



The
Fertilizer Institute
Nourish, Replenish, Grow

The History of Ammonia to 2012

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www.tfi.org

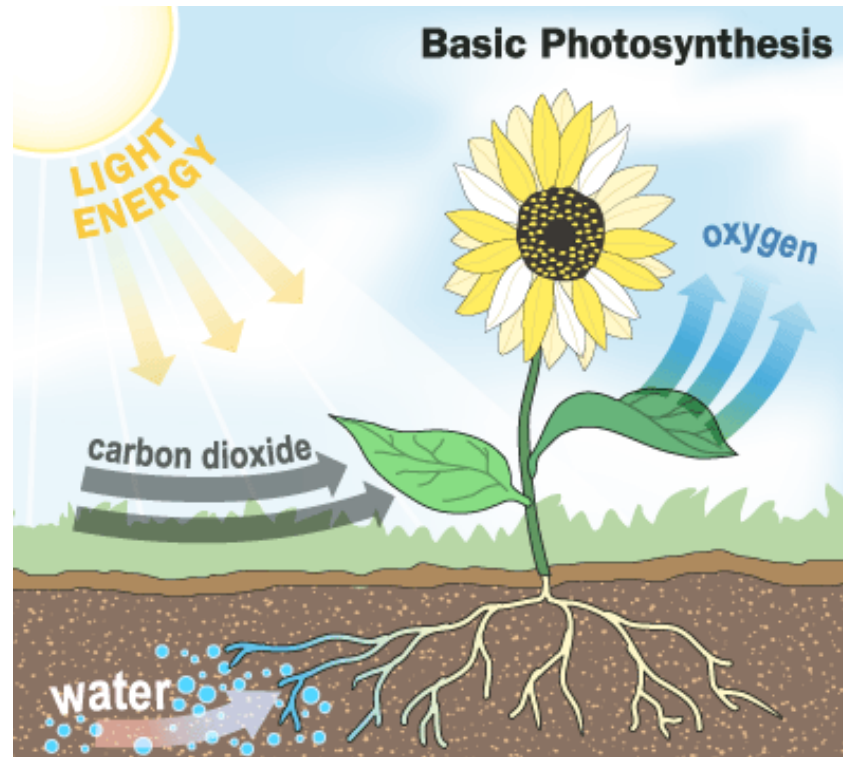
Presentation Outline

- Nitrogen Before Haber-Bosch Ammonia
- Fritz Haber, Carl Bosch, BASF and Anhydrous Ammonia
- Haber-Bosch Ammonia Production Starts
- Ammonia Production Expands
- Nitrogen Demand – Global and U.S. Profile
- Importance of Haber-Bosch N (nutrients) to mankind

1770's - Birth of Modern Chemistry



Joseph Priestly
English Minister
(1733 – 1804)



Jan Ingenhousz
Dutch Physician
(1730 – 1799)

Discovered the Fundamentals of Photosynthesis

1770's - Birth of Modern Chemistry



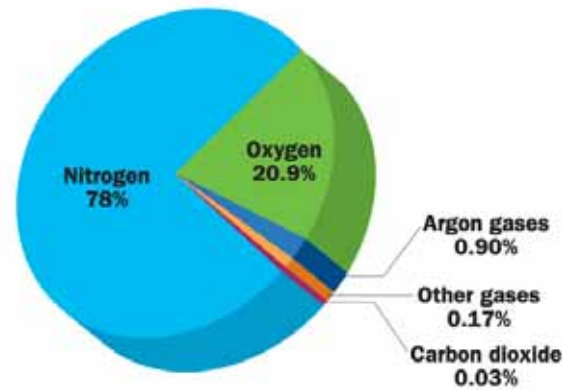
Carl Wilhelm Scheele
Swedish Chemist
(1742 – 1786)



Daniel Rutherford
English Botanist
(1749 – 1819)
"discovered" N



Antoine Laurent Lavoisier
French Chemist
(1743 – 1794)
named "azote"

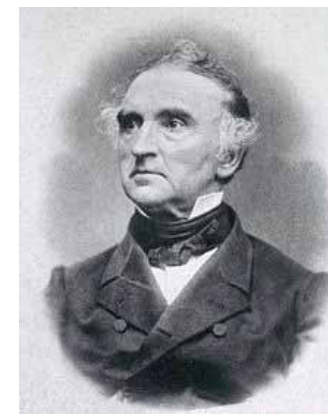


First Realization that Nitrogen Makes up Most of the Atmosphere

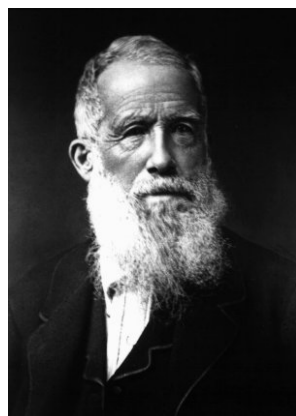


Jean-Baptists Boussingault
French Chemist
(1802 – 1887)

1800's - Value of N in Crop Production Demonstrated

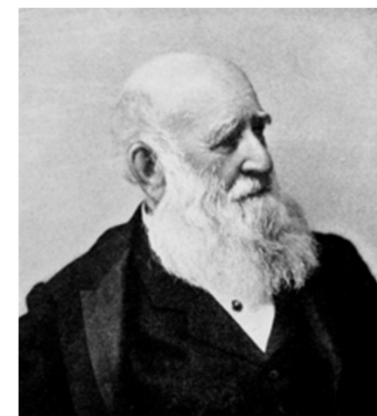


Justin von Liebig
German Chemist
(1803 – 1873)



John Bennet Lawes
English Scientist
(1814 – 1900)

Scientific Experiments Left No Doubt About Nitrogen's Crucial Role in Crop Production!



Joseph Henry Gilbert
English Chemist
(1817 – 1901)

Birth of Anhydrous Ammonia – NH₃

Claude-Louis Berthollet
French Chemist
(1748 – 1822)

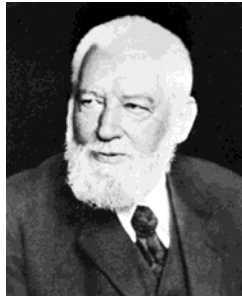
1784 – Berthollet becomes aware that the element “azote” joins with hydrogen to form ammonia

Georg Friedrich
Hildebrand
German Professor
(1764 - 1816)

1795 – The first failed attempt to combine N and H

More than a
dozen chemists

1800’s – Conducted experiments to synthesize ammonia



Wilhelm Ostwald
German Chemist
(1853 – 1932)

**1900 – Ostwald thought he had succeeded in synthesizing ammonia
– Tests by Carl Bosch for BASF proved him wrong;
NH₃ production was result of contaminants in machine**

Nitrogen Sources: Pre Haber-Bosch

Manures (barnyard and other)

Guano (solidified bird excrement accumulated on subtrop/tropical islands)

Chilean Nitrate (sodium nitrate)

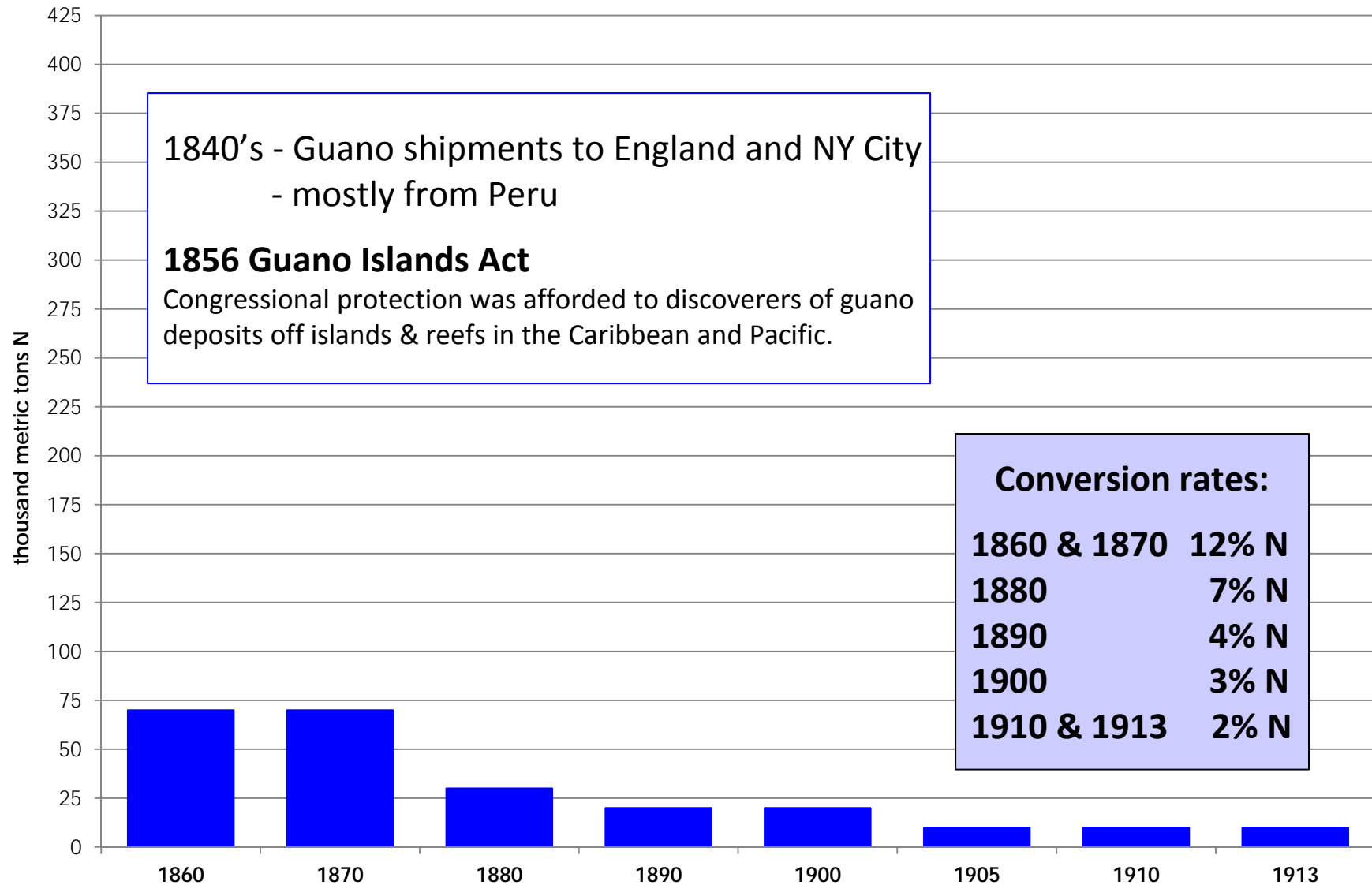
Coke-Oven Gas (by-product) Ammonium Sulfate

- Coke oven gases produced primarily in the steel industry
- Produced by reacting recovered coke oven ammonia with sulfuric acid

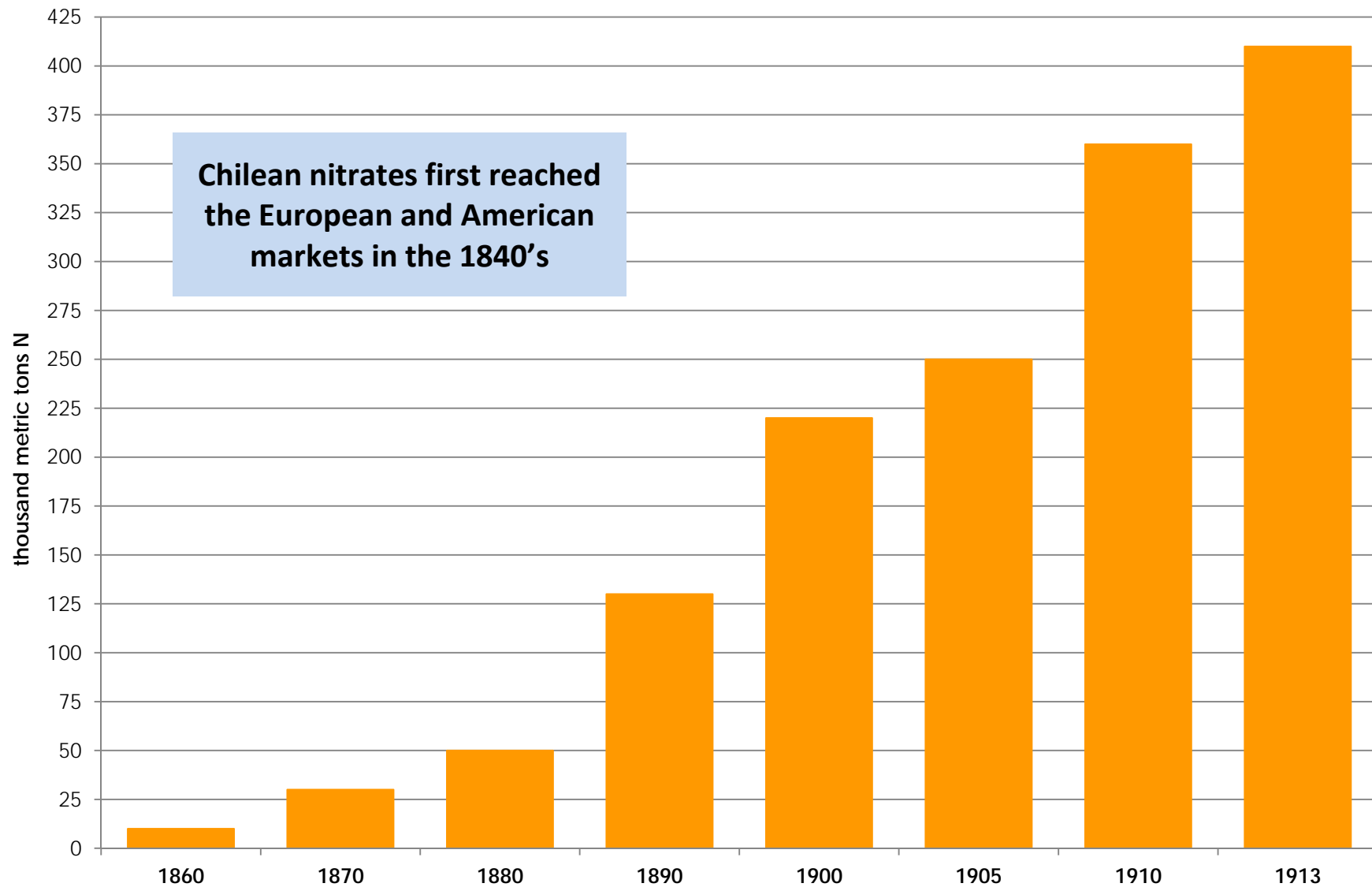
Nitrogen Fixation Processes:

- **Electric Arc Process (1901 – Niagara Falls)**
Uses electrical current to combine N and oxygen in the air to form nitric oxide (NO)
(NO) → Nitric Acid → Calcium Nitrate/Sodium Nitrate
- **Cyanamide Process (1907 – Italy)**
Limestone burned to form calcium oxide; fused with coke in electric furnace → calcium carbide
Calcium carbide (CaC_2) reacted with N at high temperature to form calcium cyanamide (CaCN_2)
Calcium cyanamide (CaCN_2) → ammonia; ammonium phosphates; nitric acid; ammonium nitrate

N Production as Guano Extraction, Pre Haber-Bosch

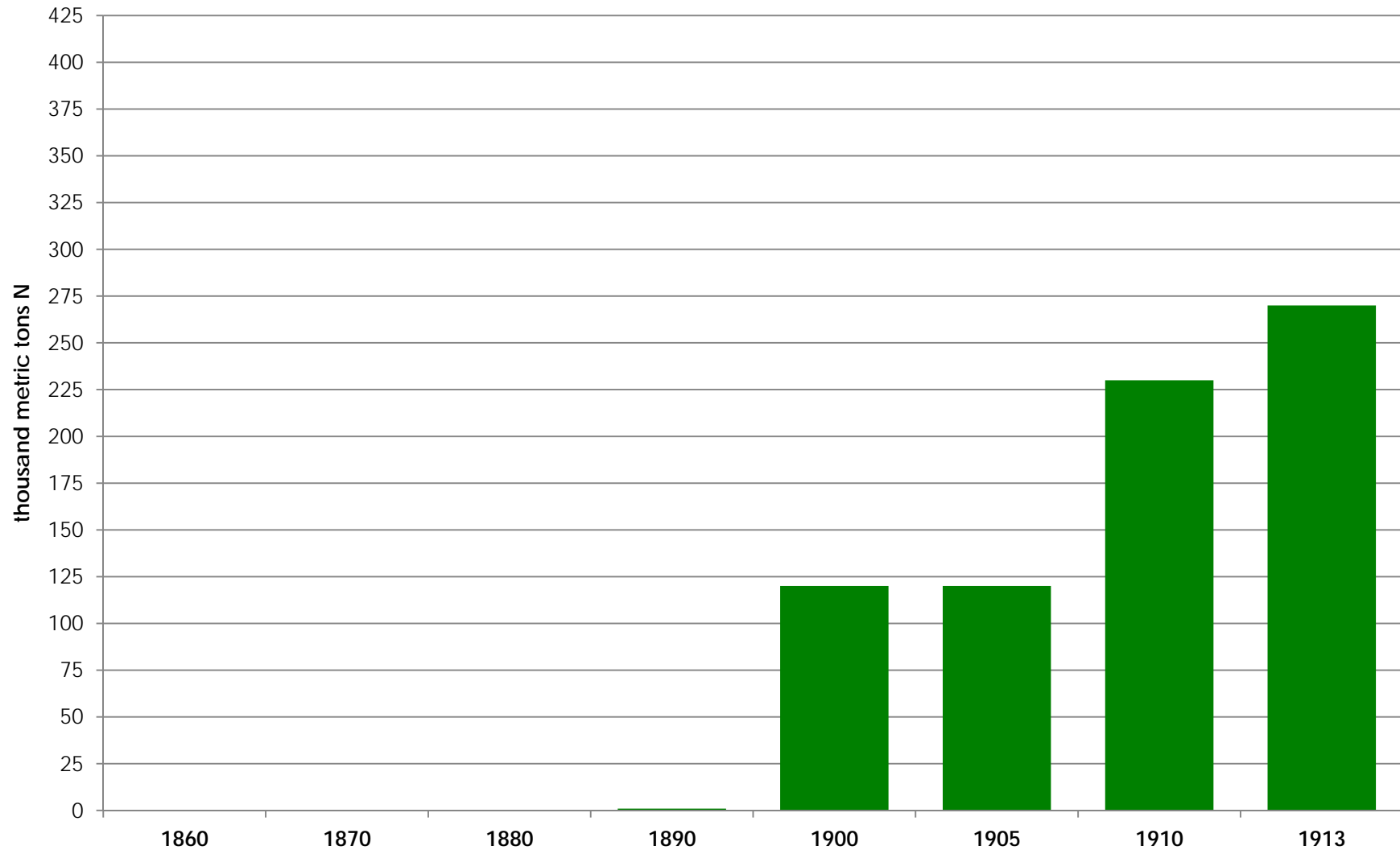


Chilean Nitrate Production, Pre Haber-Bosch

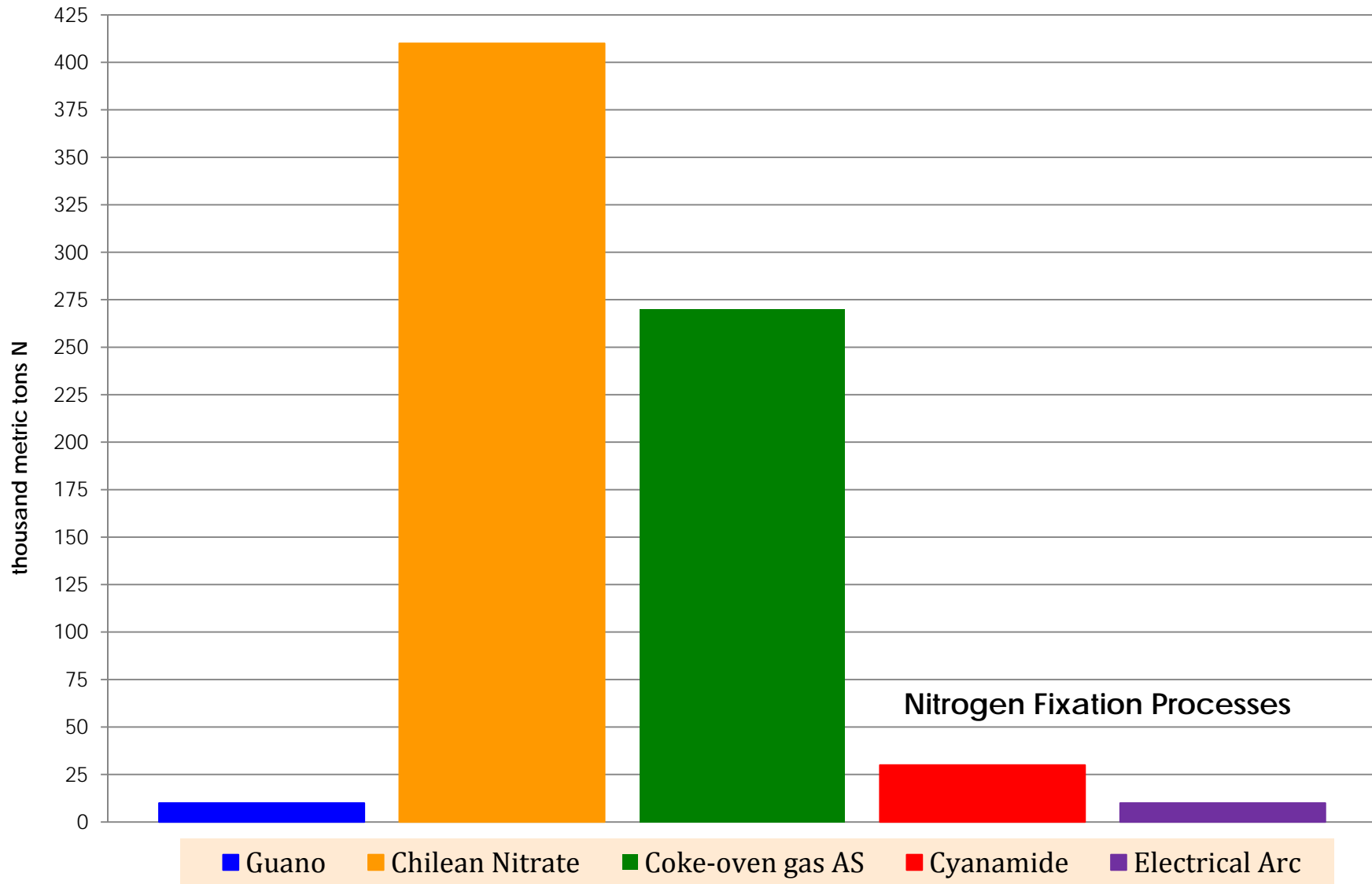


Coke-oven Gas (by-product) AS

Production, Pre Haber-Bosch



Pre Haber-Bosch N Production, Excluding Ammonia - 1913



Evaluating Haber's Machine



Heinrich von Brunck
CEO of BASF



August Bernthsen
BASF Director of Research



Carl Bosch
BASF Chemist



Fritz Haber

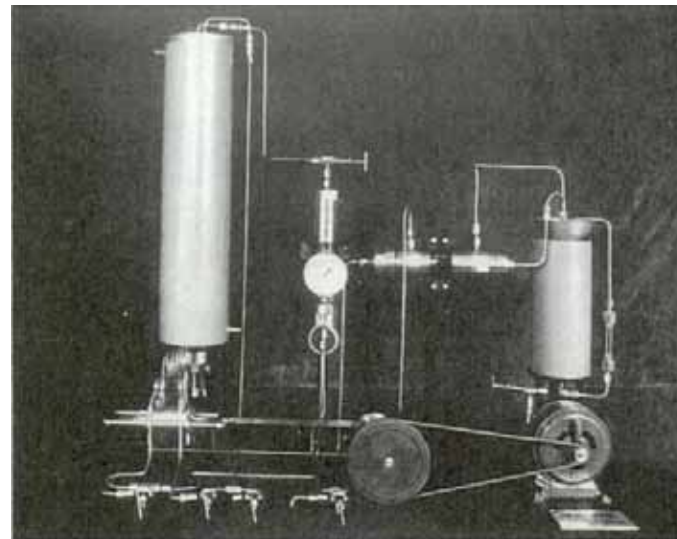
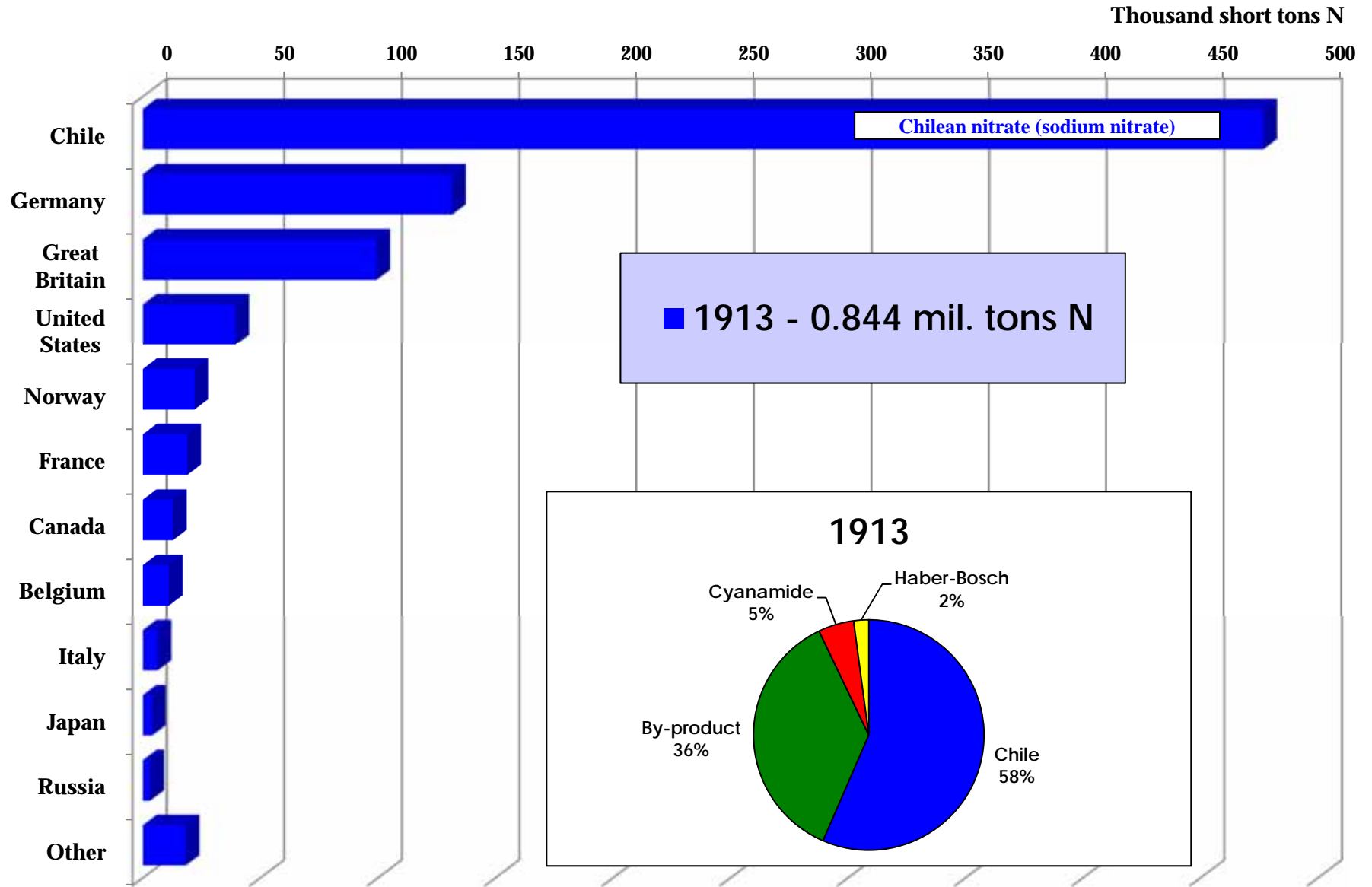


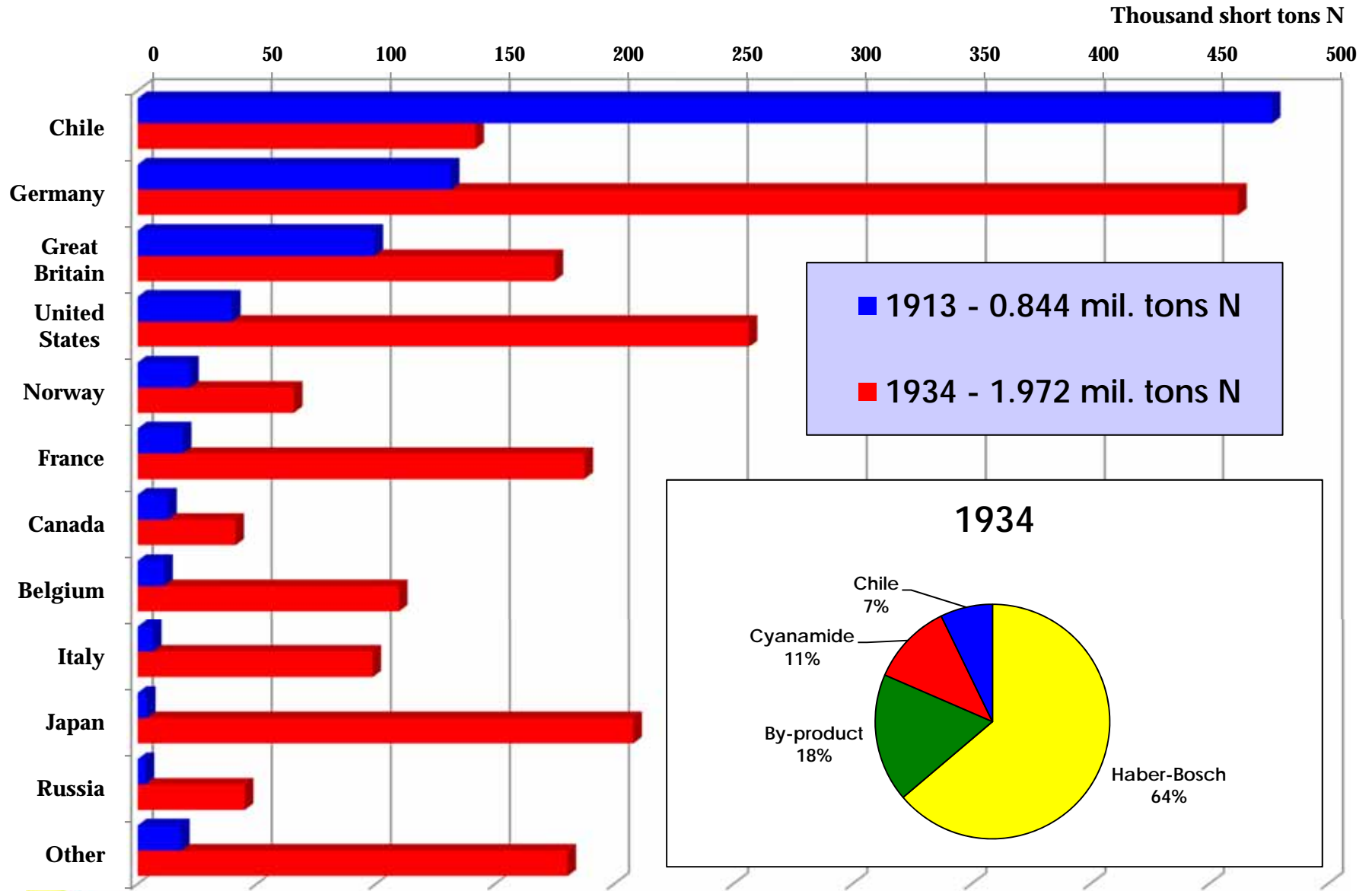
Abb. 4 Haber-Le Rossignol-Apparatur zur Ammoniaksynthese



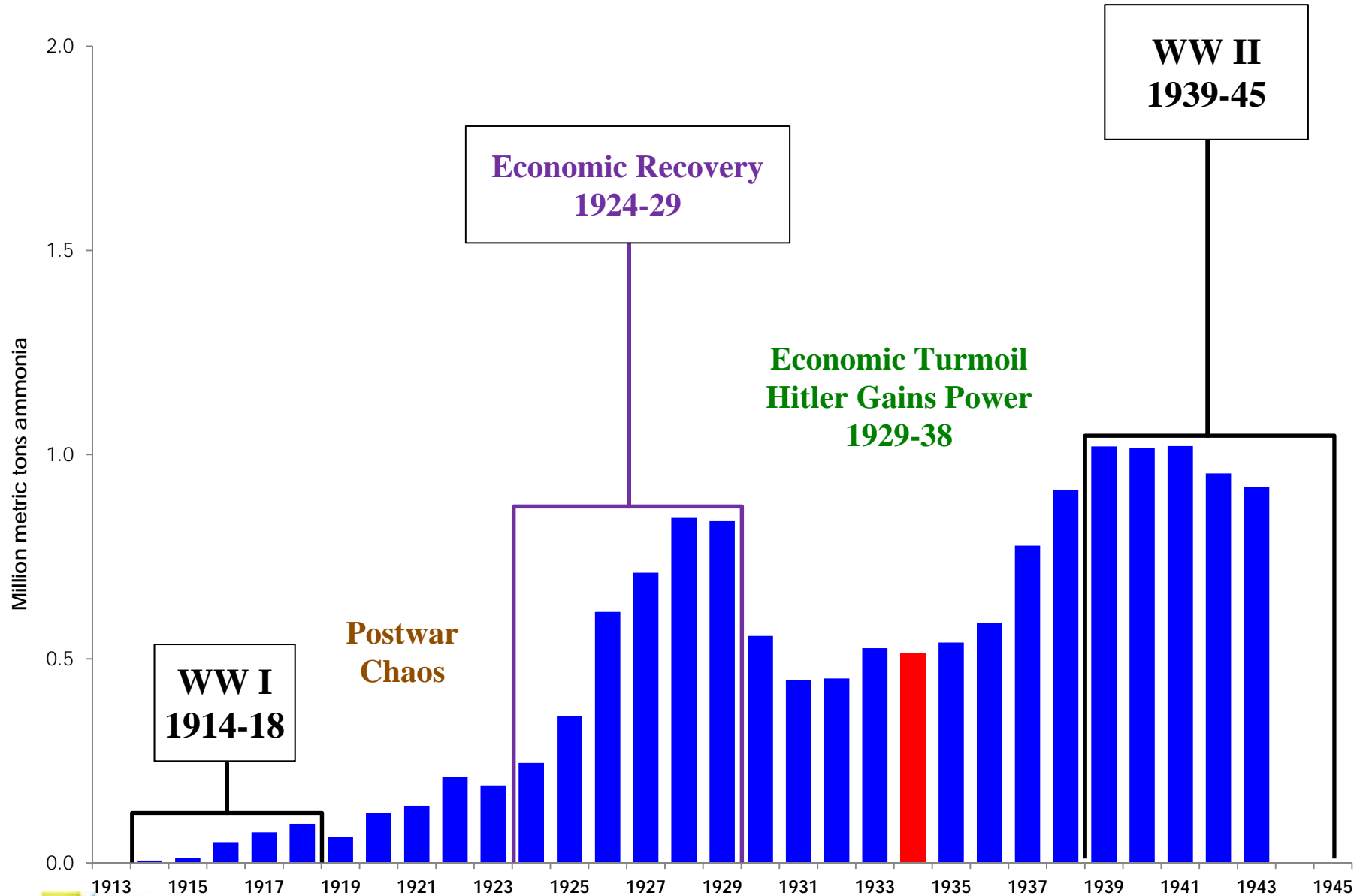
Global N Production



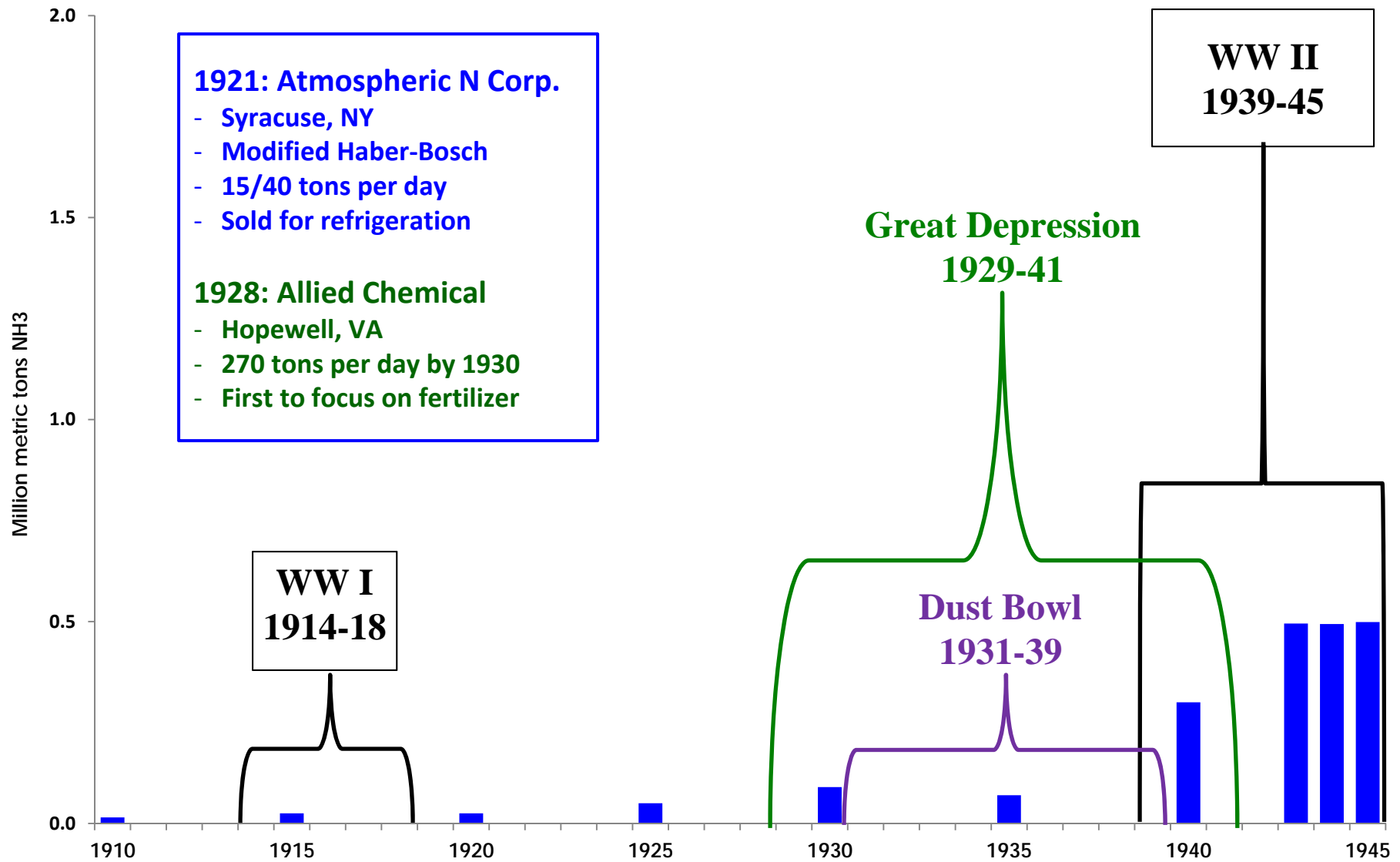
Global N Production



German Haber-Bosch Ammonia Production

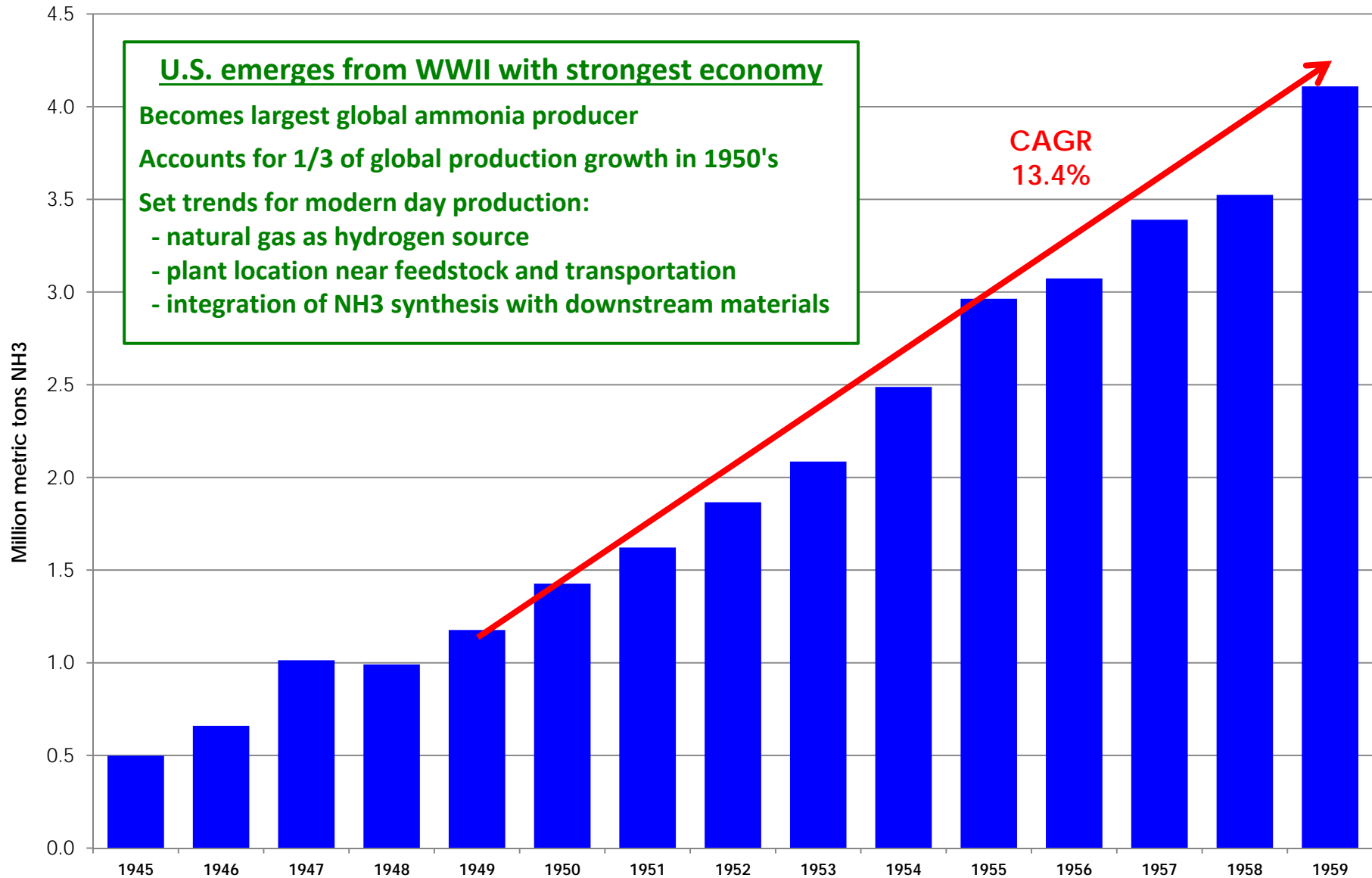


U.S. Ammonia Production

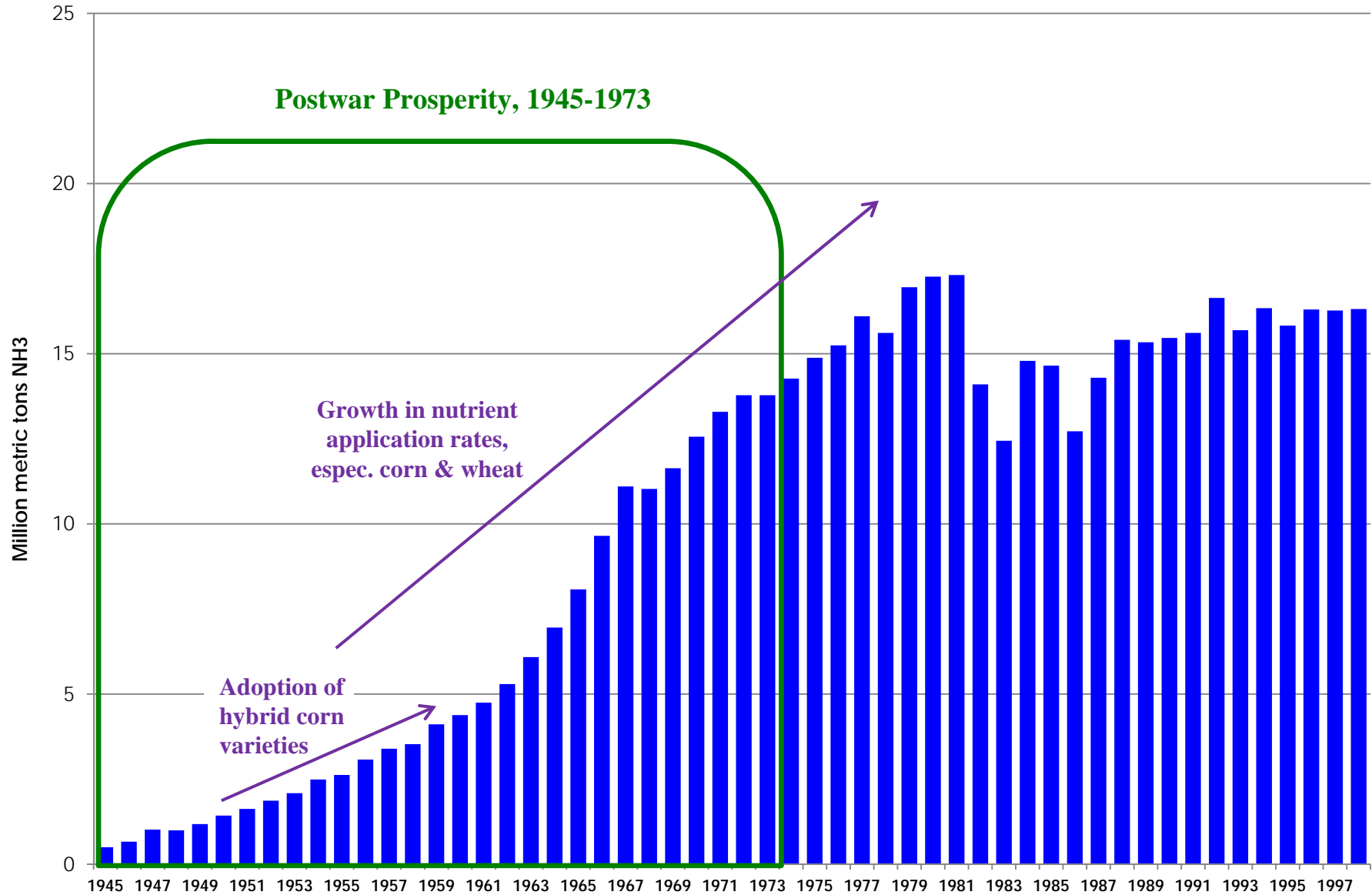


Source: Nelson, 1990; Data: 1910-1940 - Smil, 2001; 1943-1945 - U.S. Geological Survey.

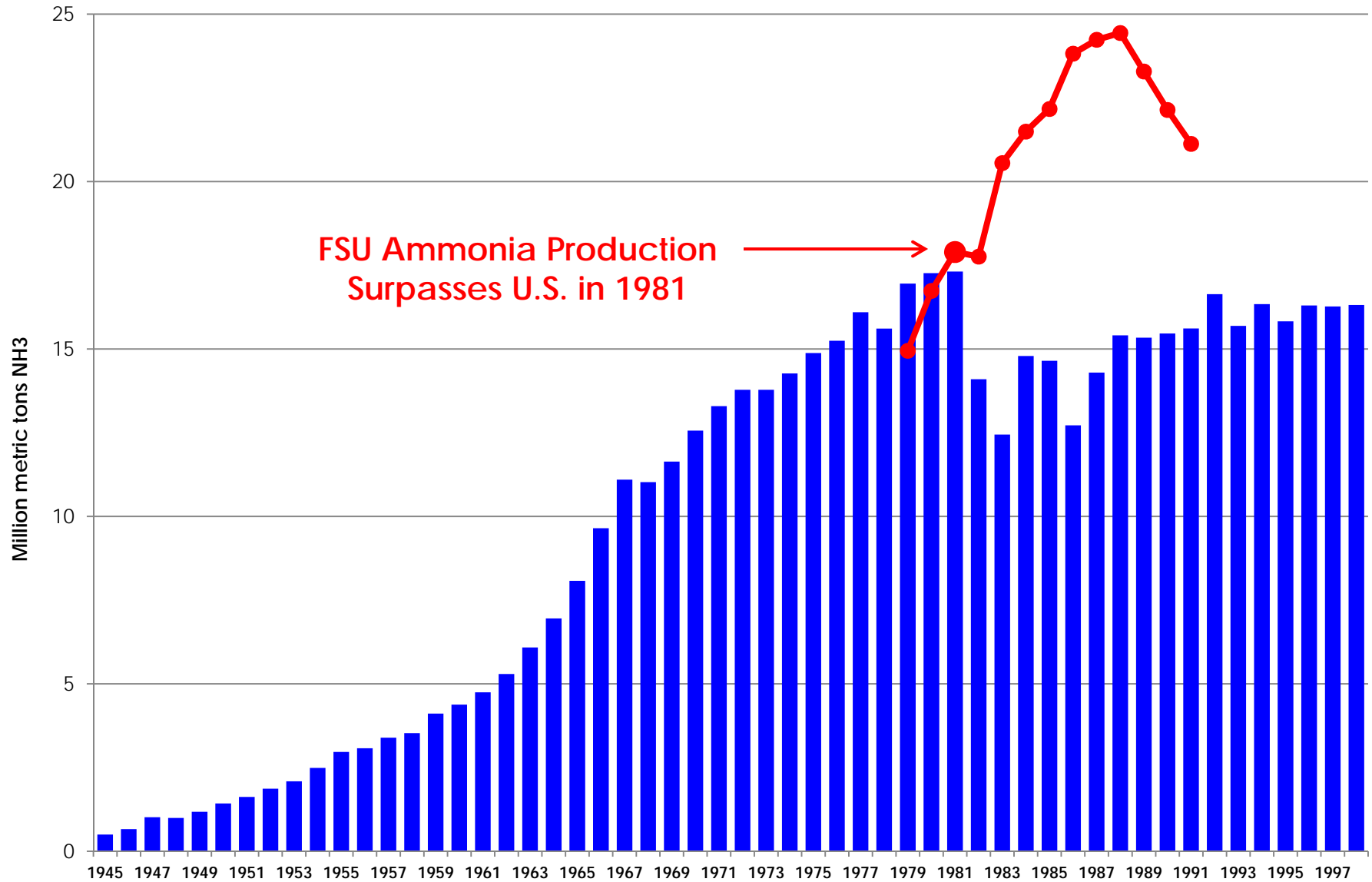
U.S. Ammonia Production



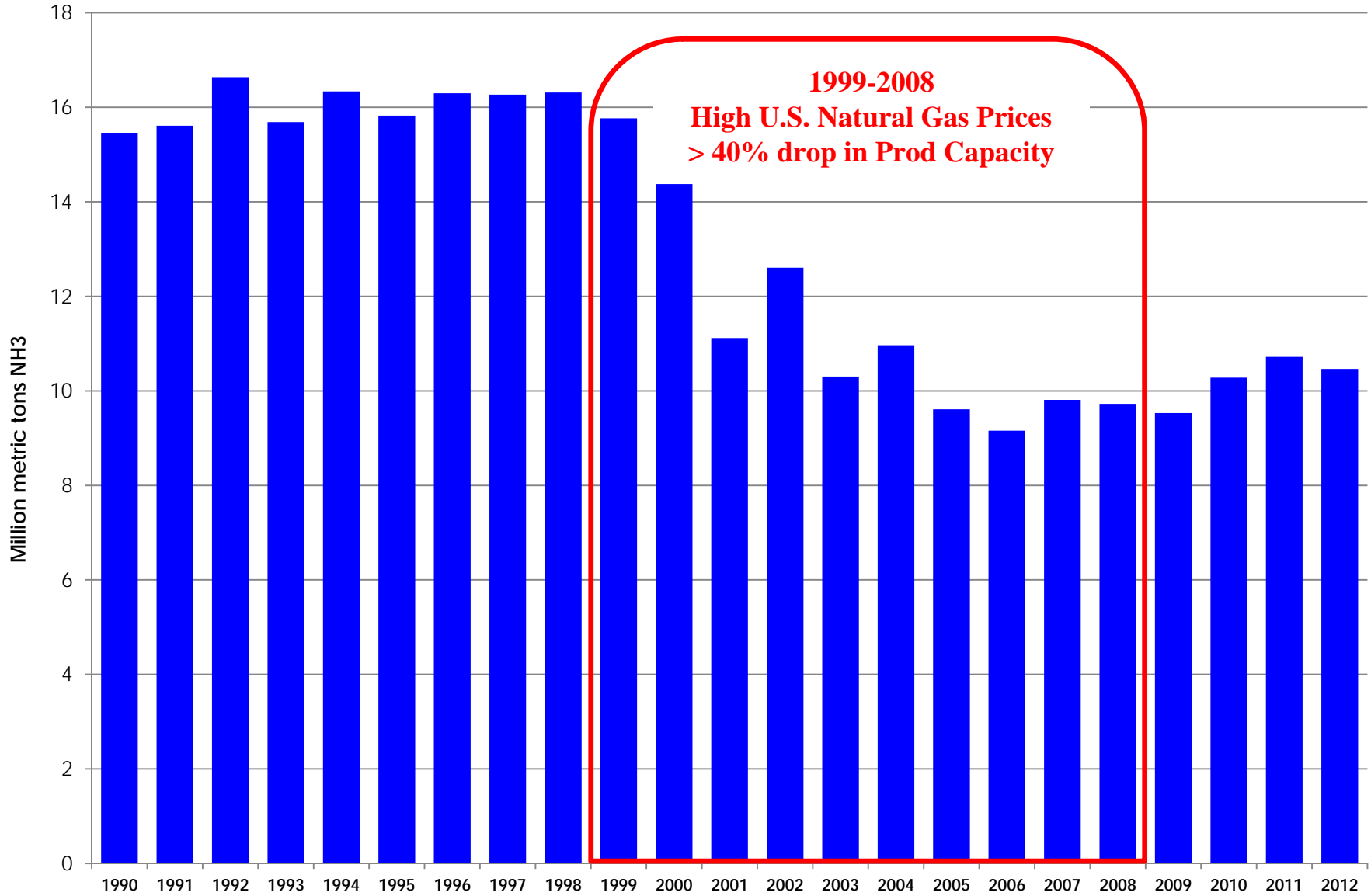
U.S. Ammonia Production, 1945-1998



U.S. Ammonia Production, 1945-1998



U.S. Ammonia Production

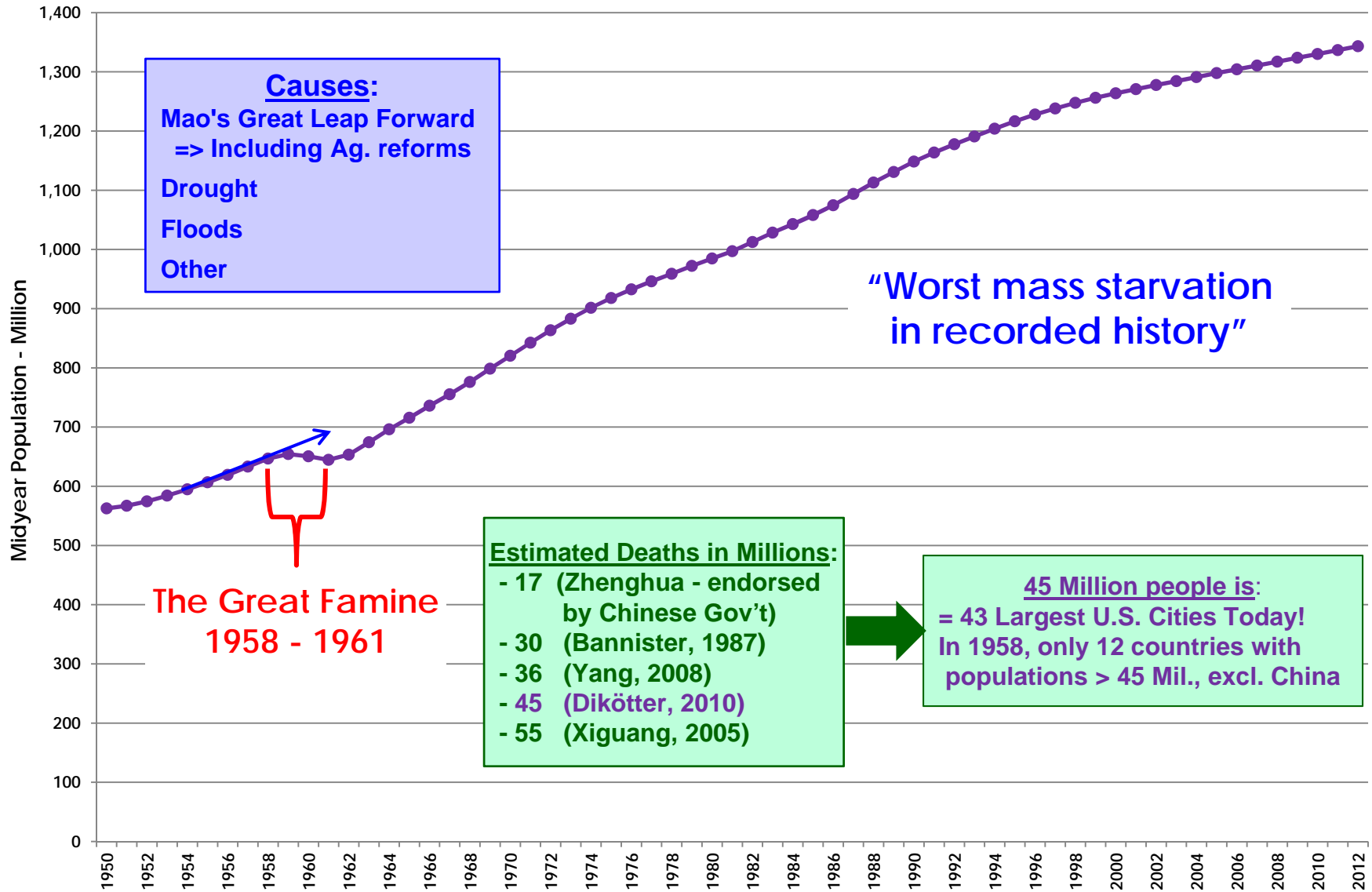


CHINA

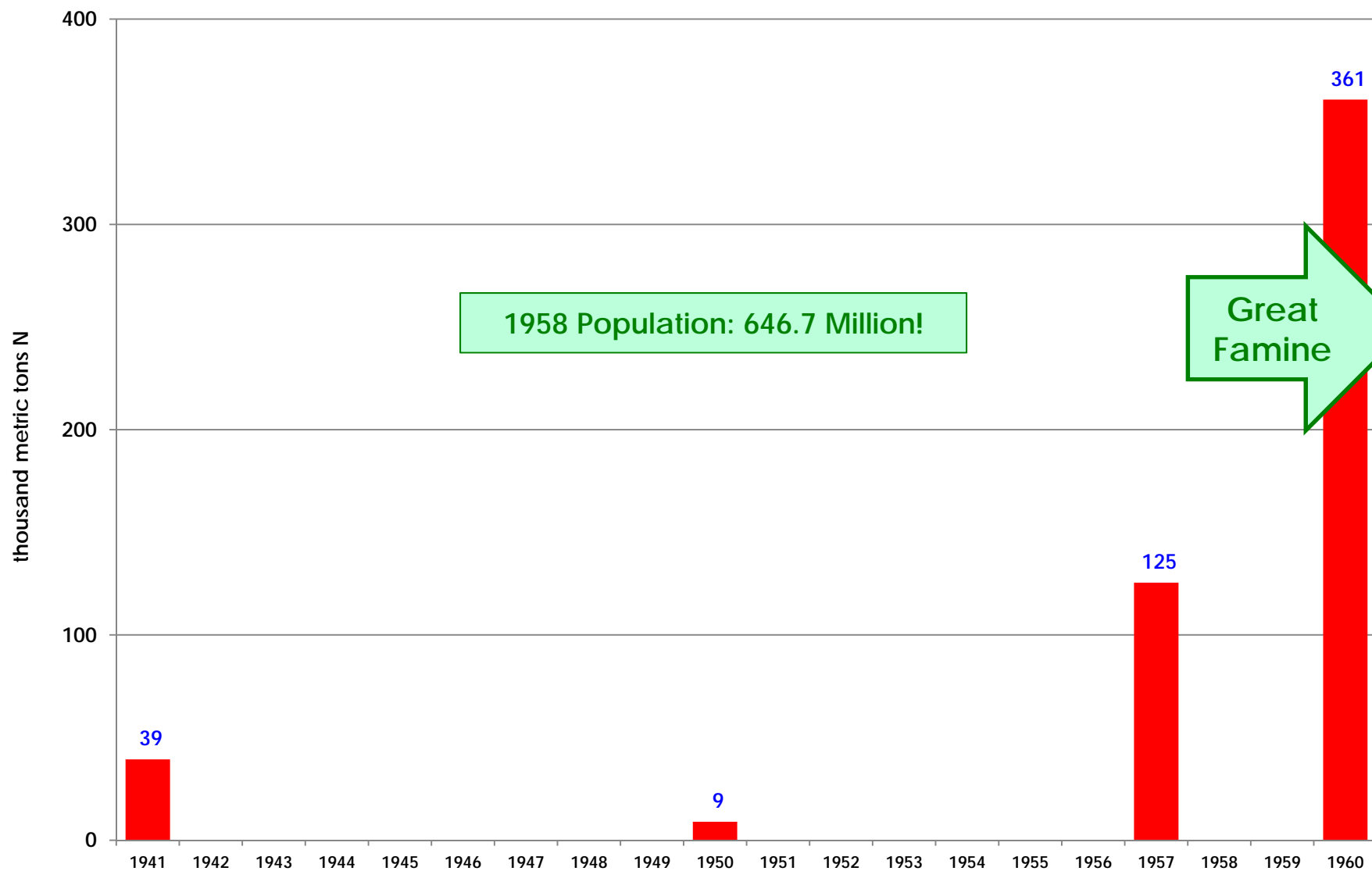
隨著食品在手，你安心

“With food in hand, you
have peace of mind”

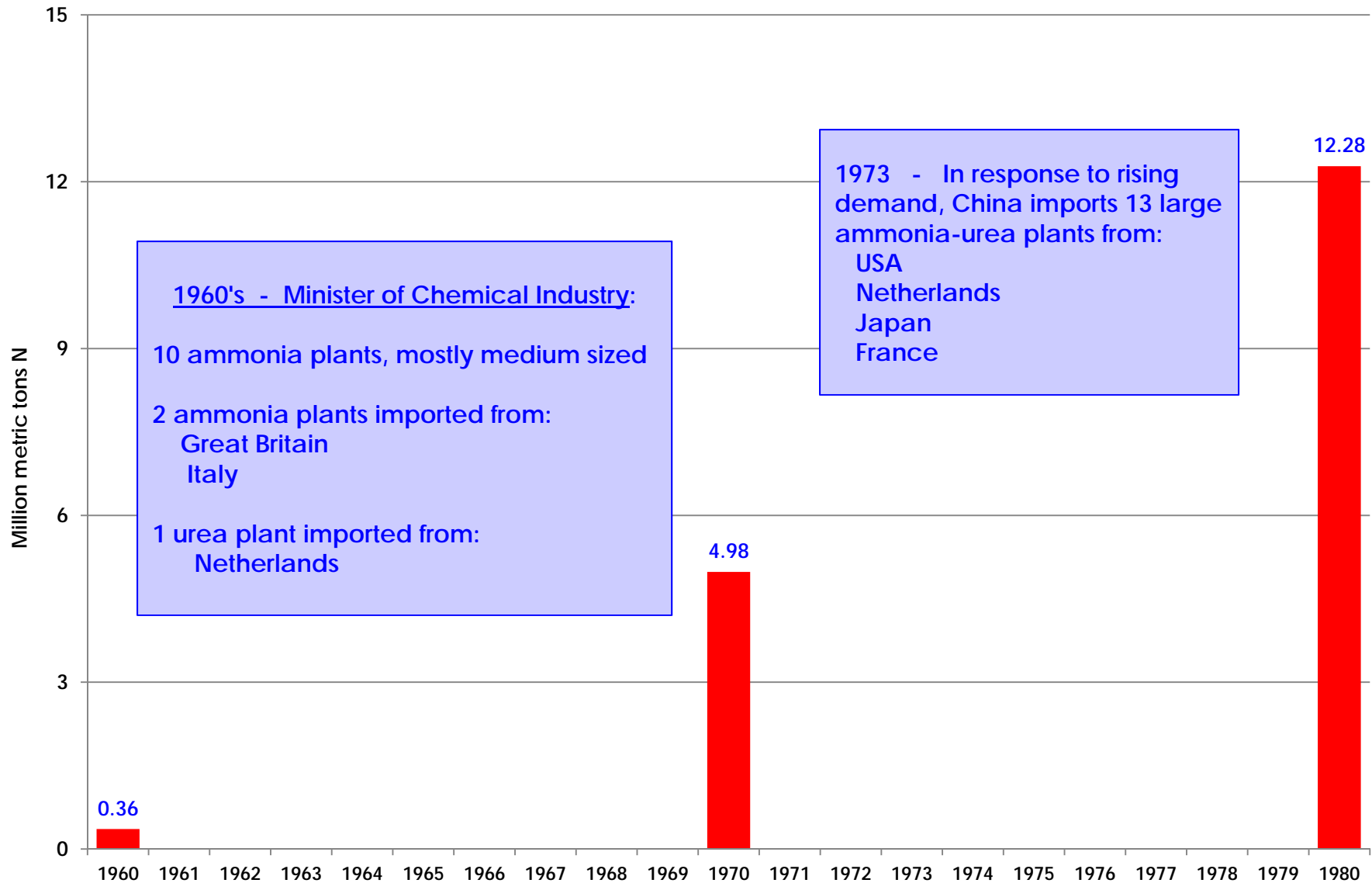
China – The Great Famine



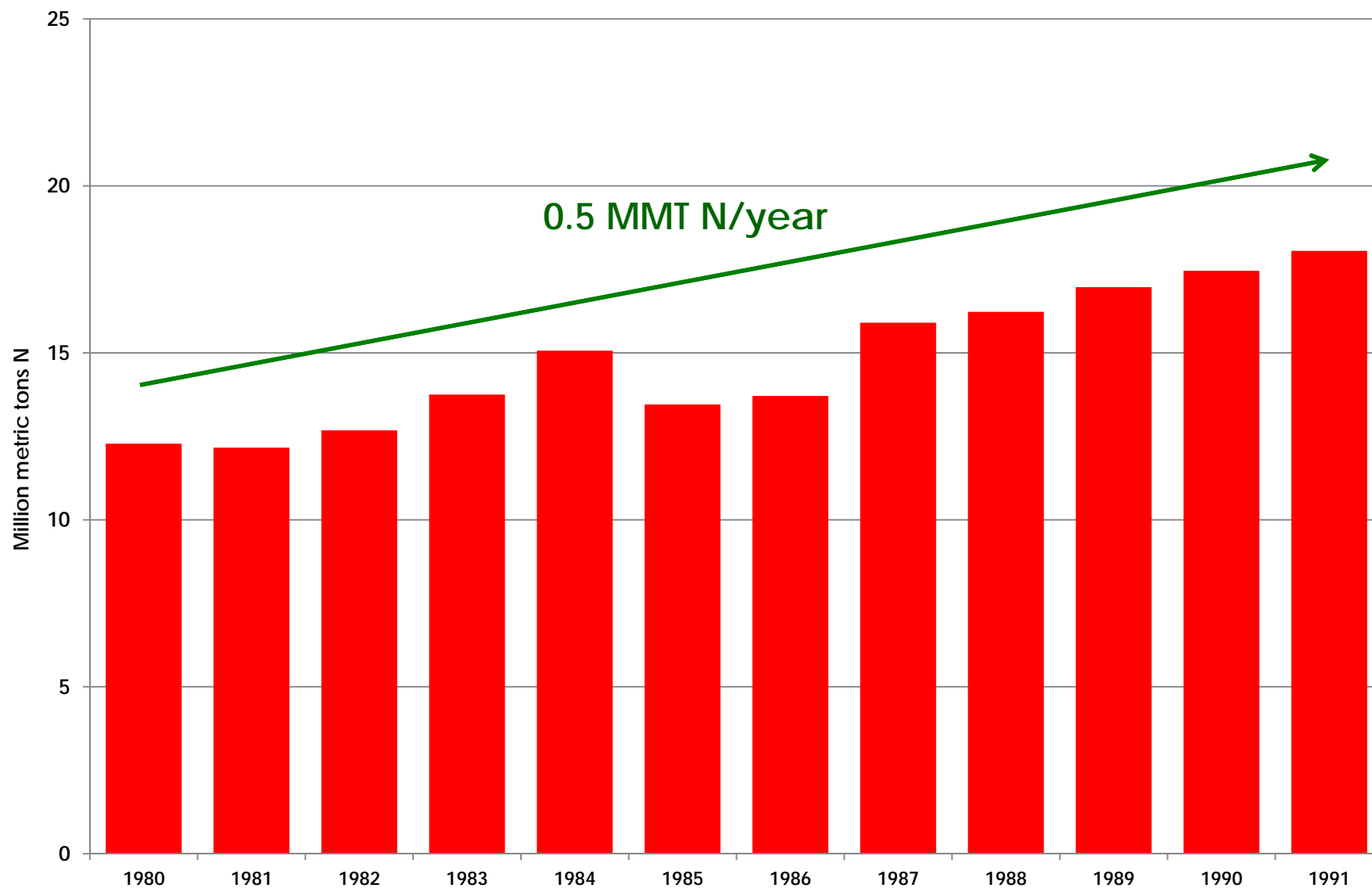
China Ammonia Production, 1941-1960



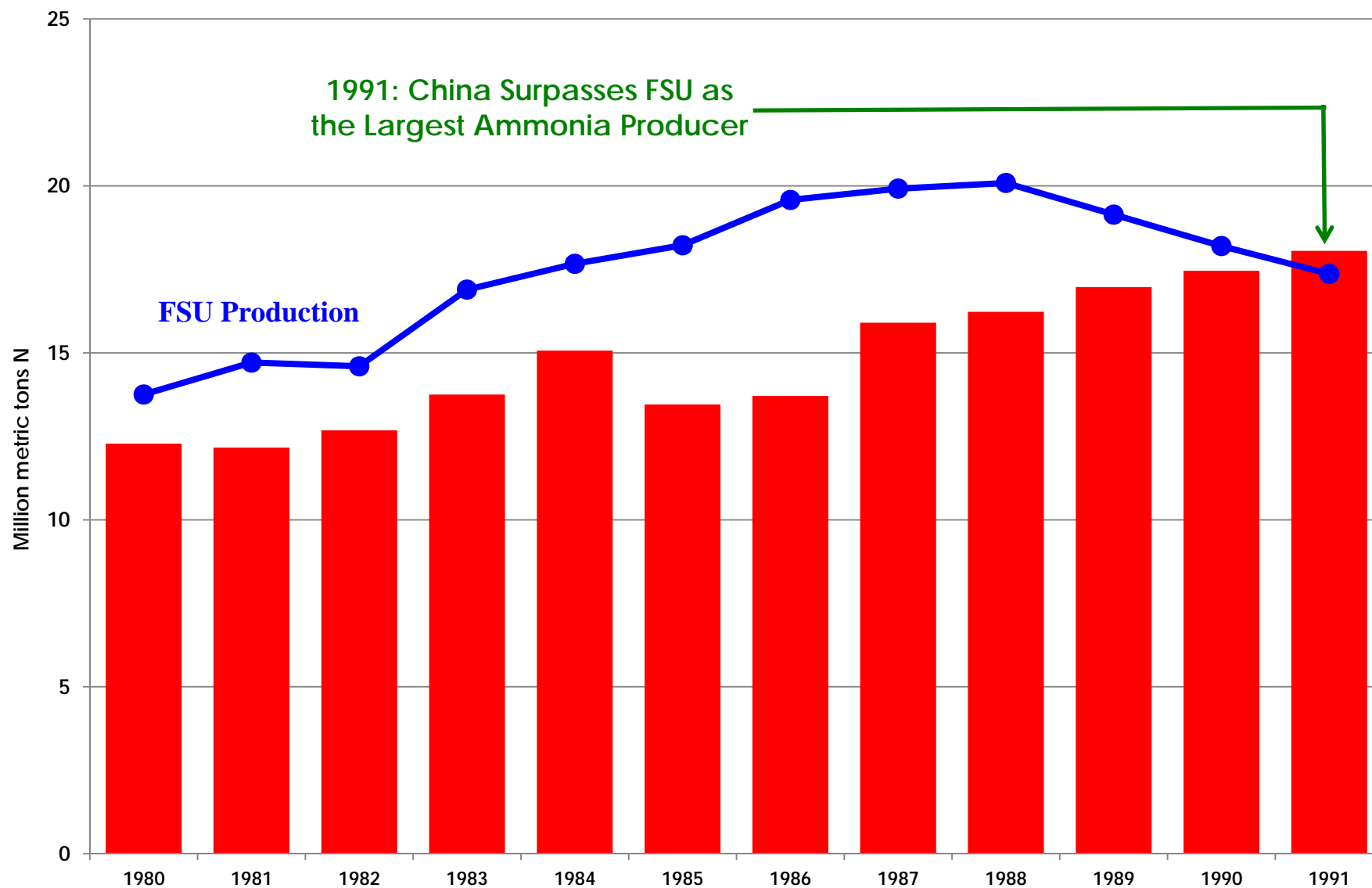
China Ammonia Production, 1960-1980



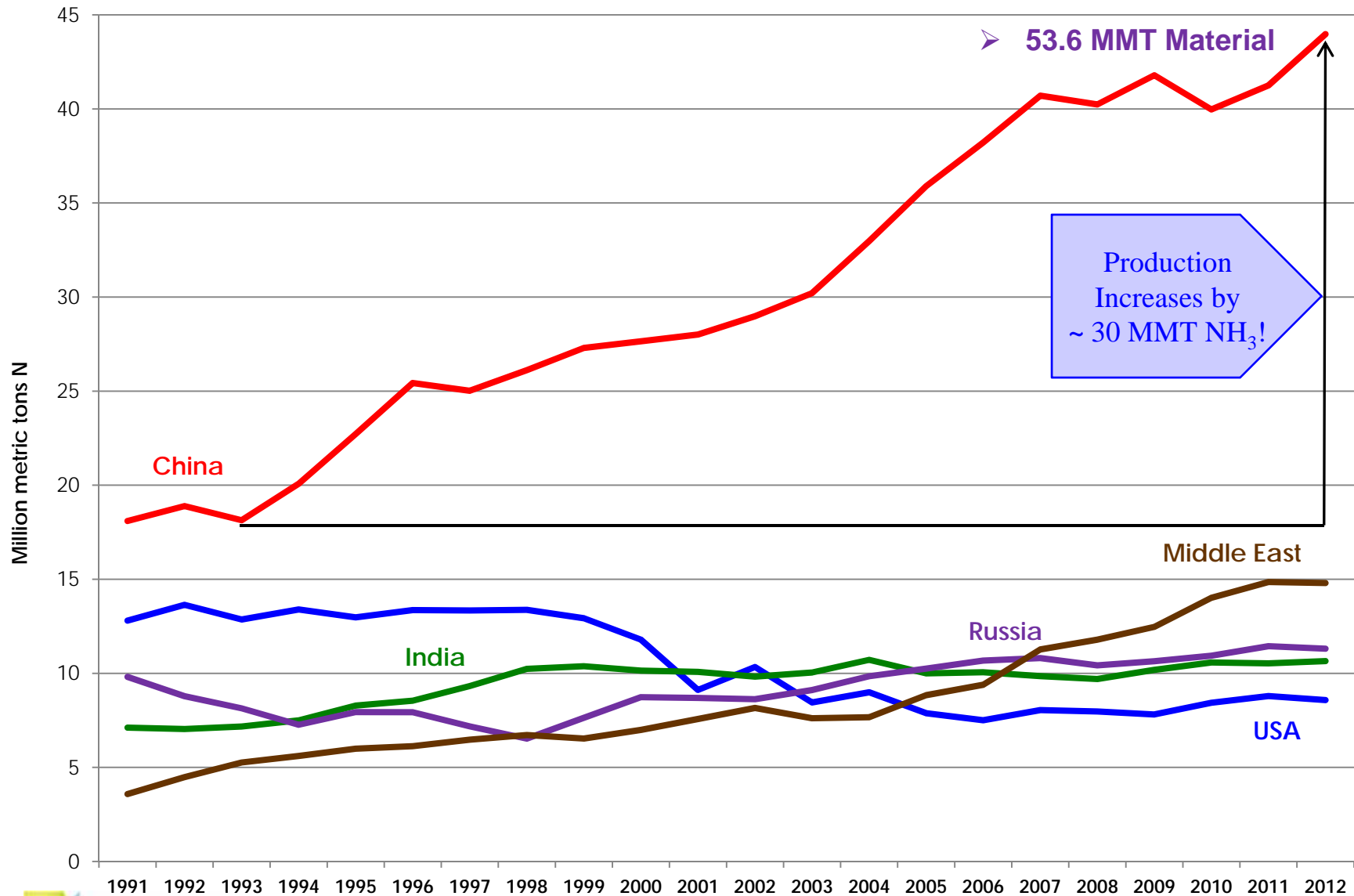
China Ammonia Production, 1980-1991



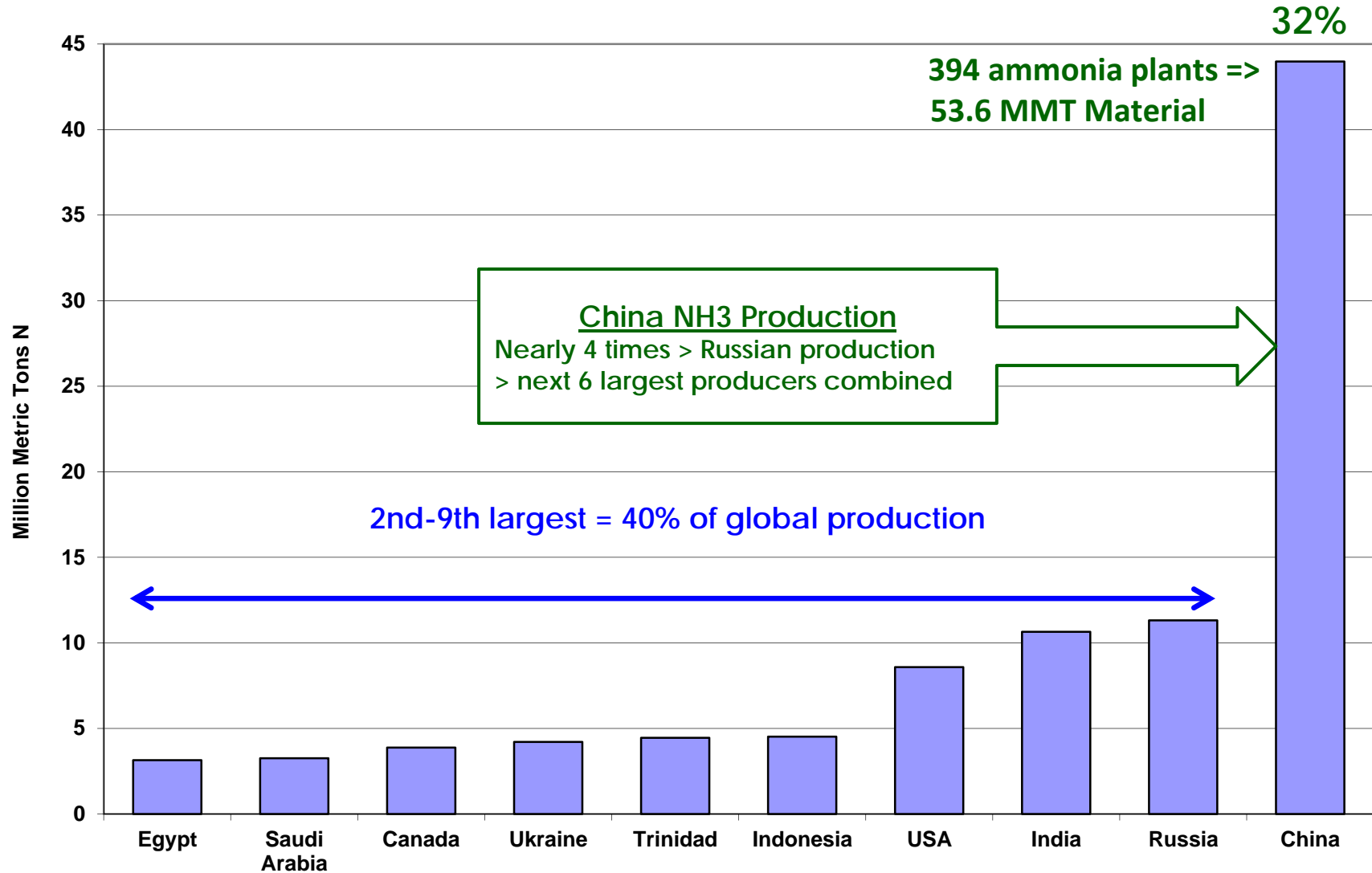
China Ammonia Production, 1980-1991



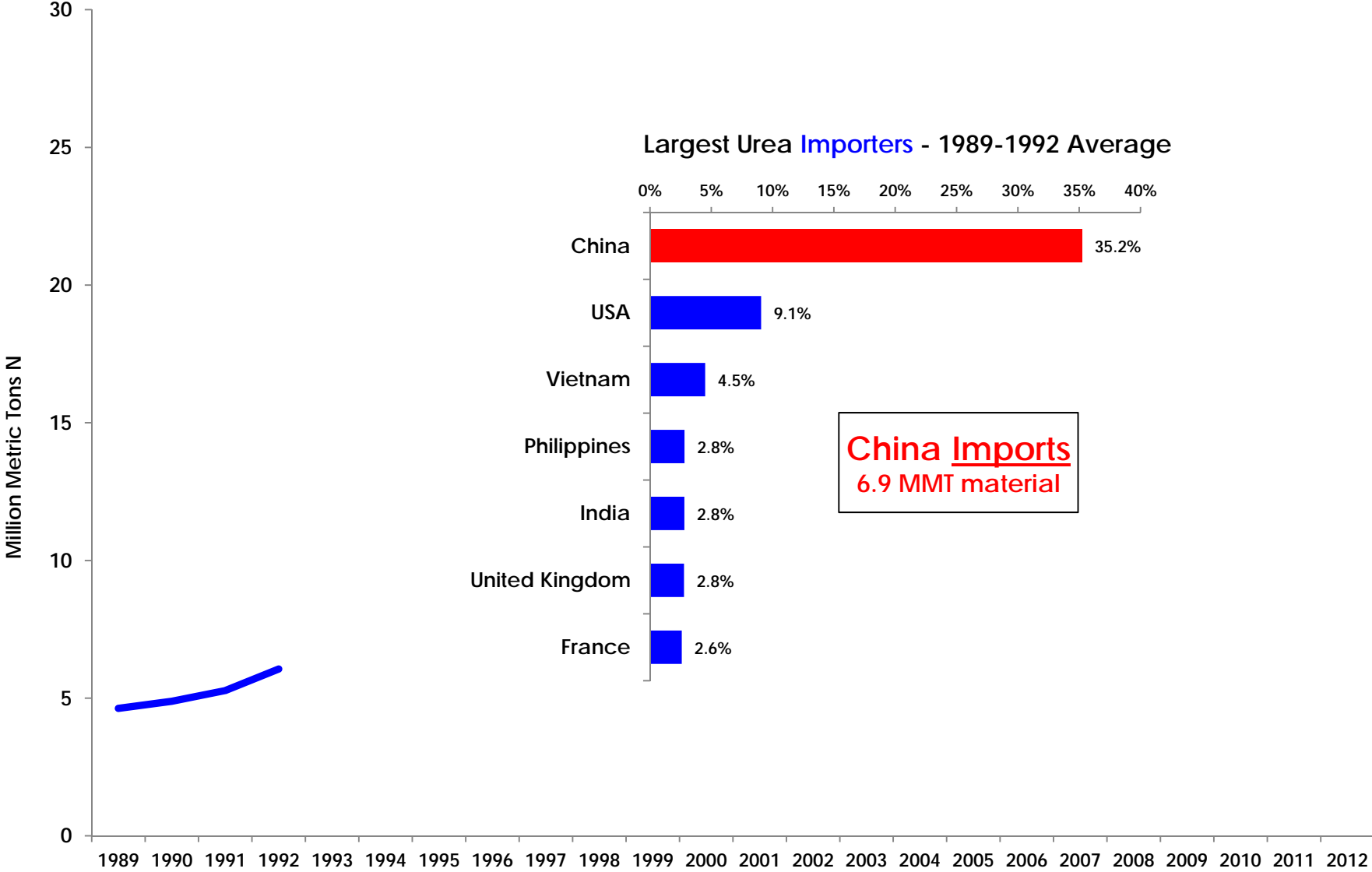
Largest Ammonia Producing Countries



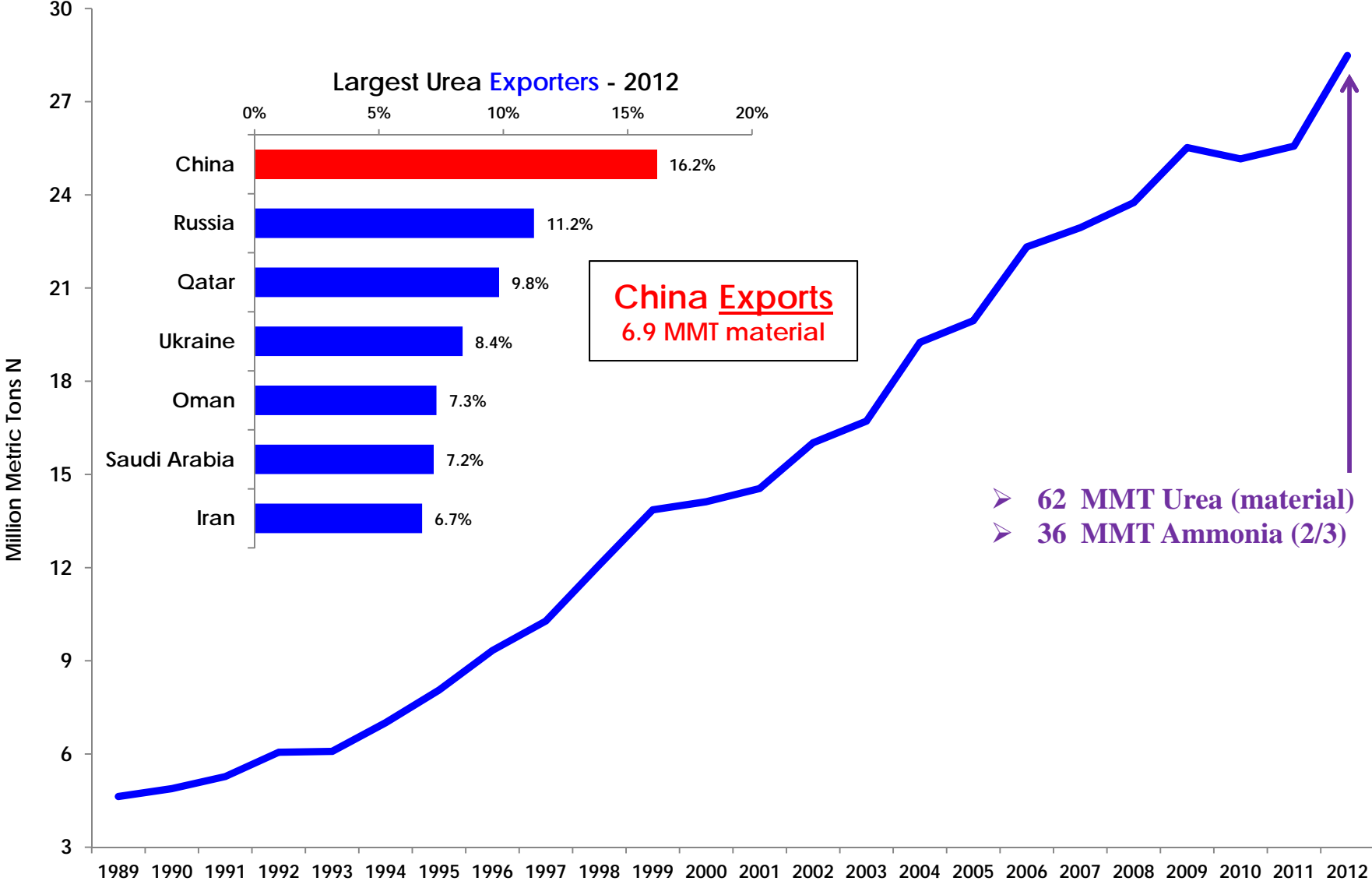
Top 10 Ammonia Producing Countries in 2012



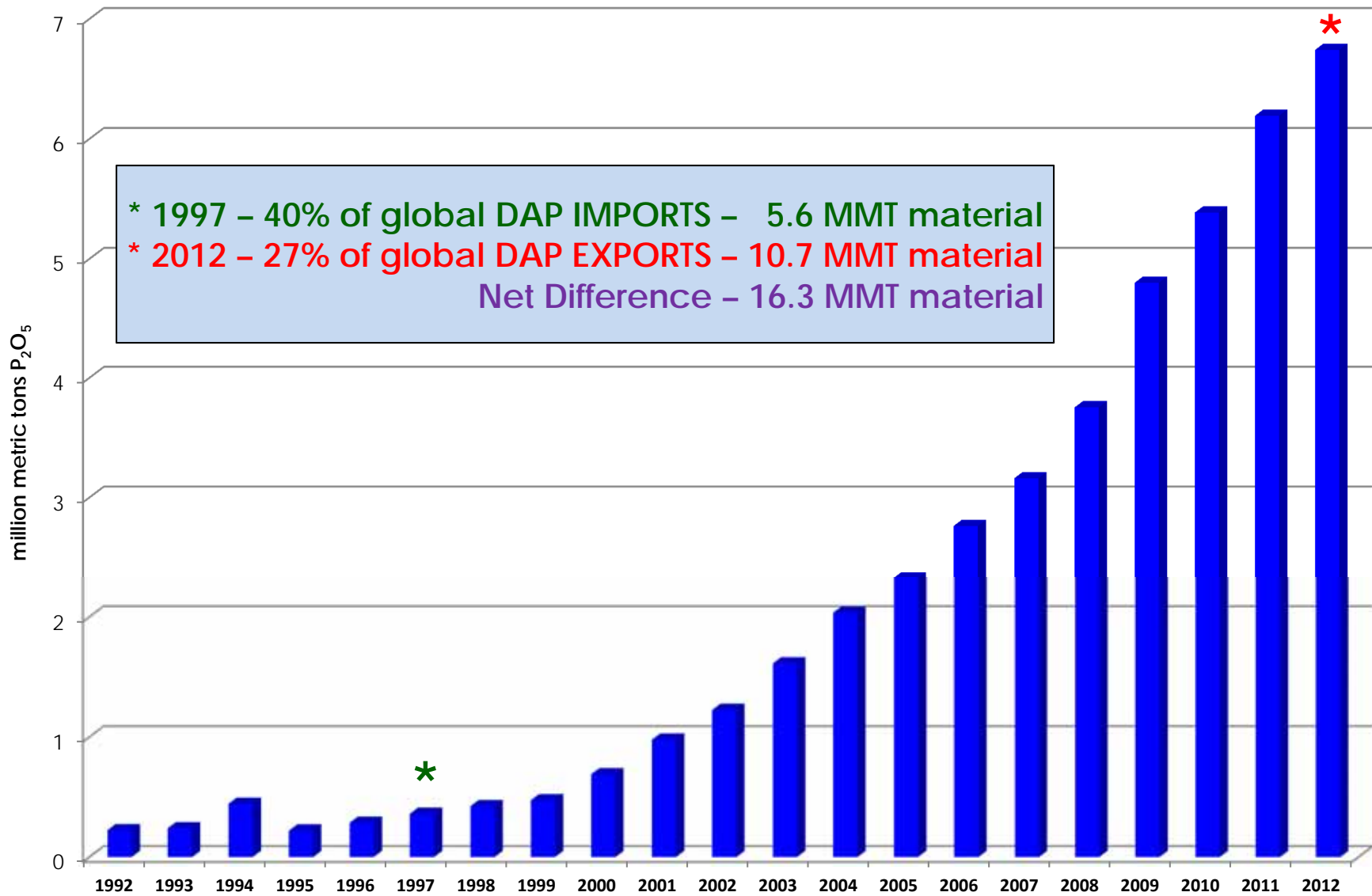
China - Urea Production



China - Urea Production

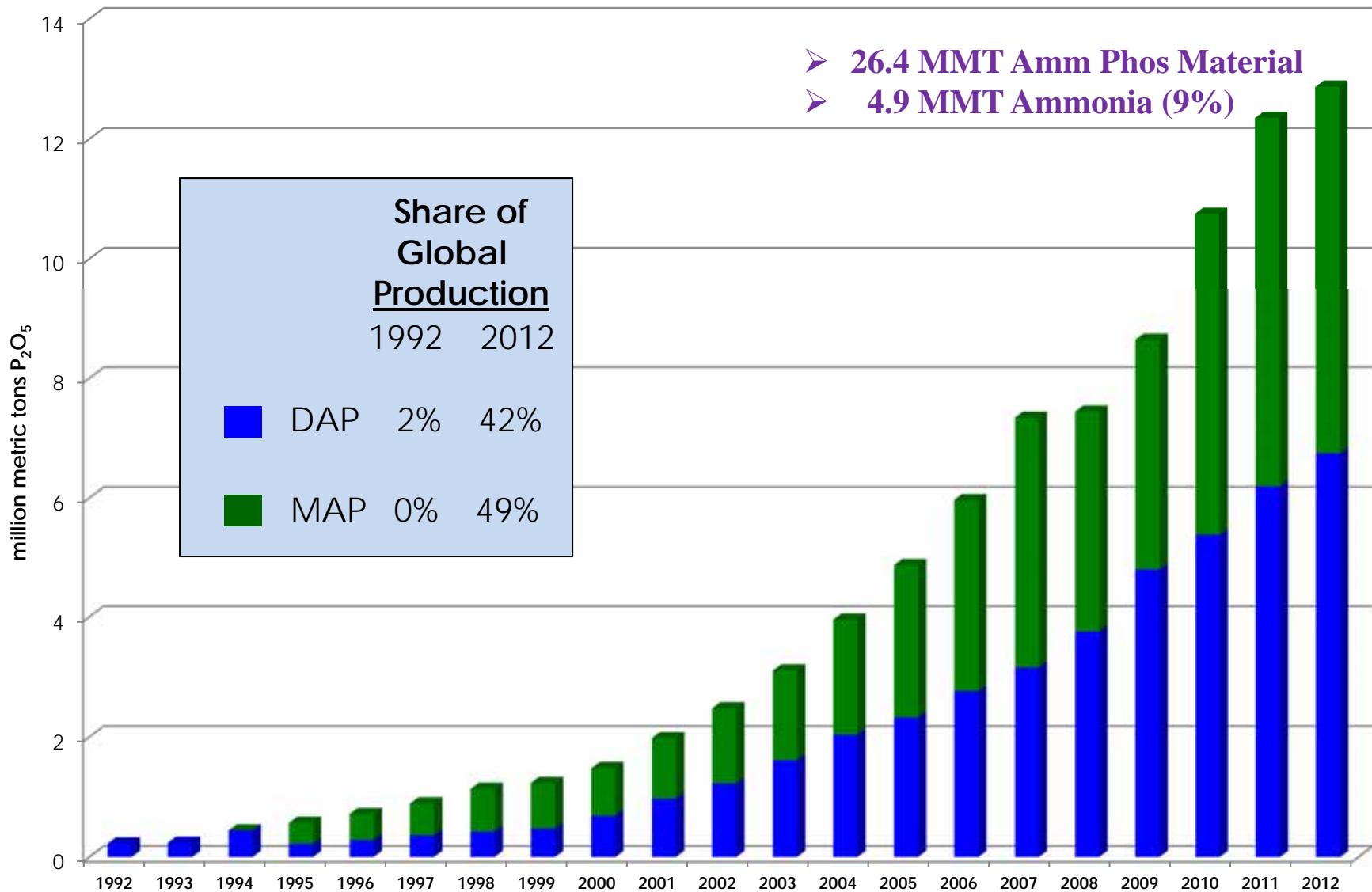


China: DAP Production

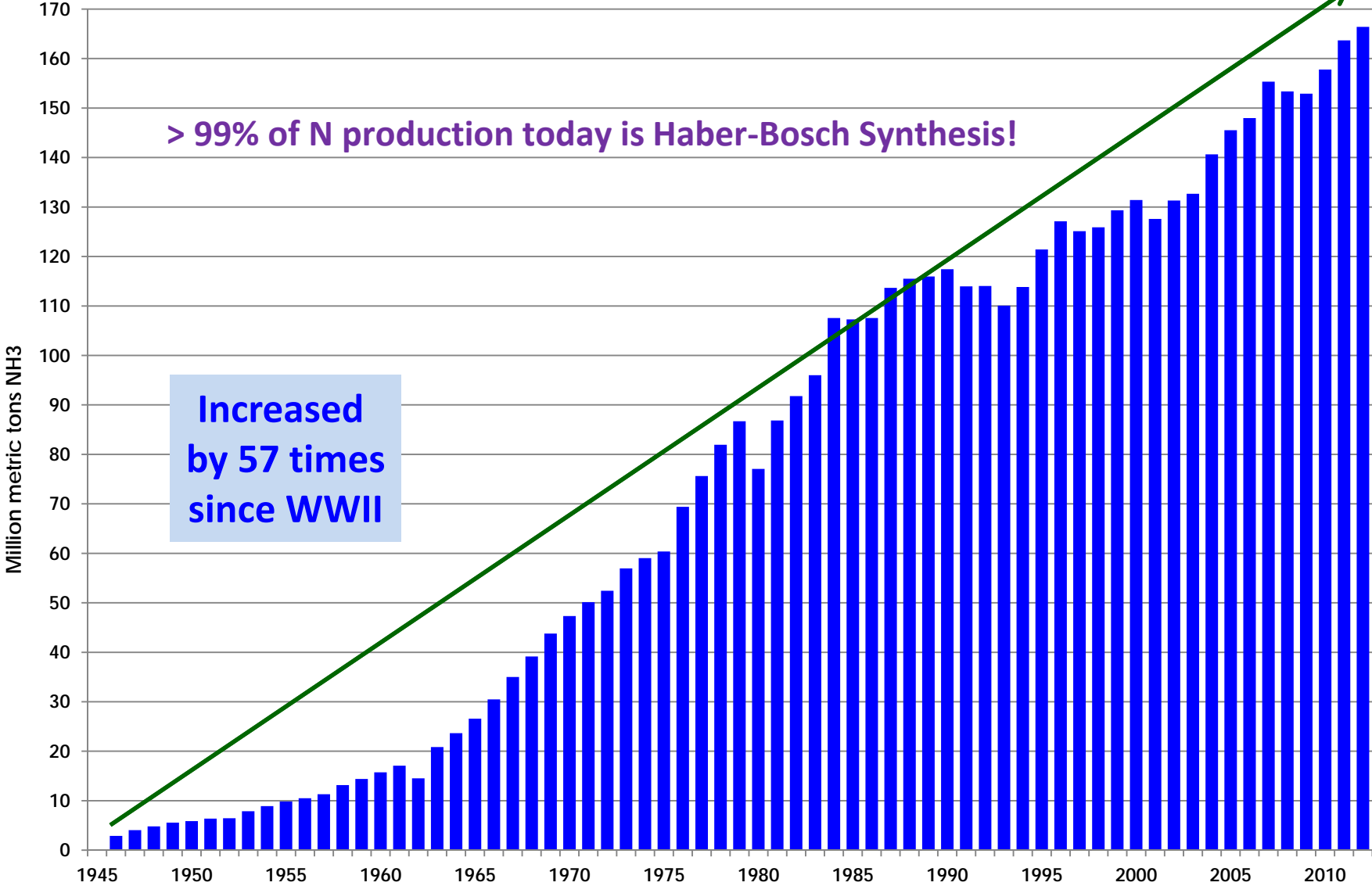


* 1997 – 40% of global DAP IMPORTS – 5.6 MMT material
* 2012 – 27% of global DAP EXPORTS – 10.7 MMT material
Net Difference – 16.3 MMT material

China: Phosphate Fertilizer Production

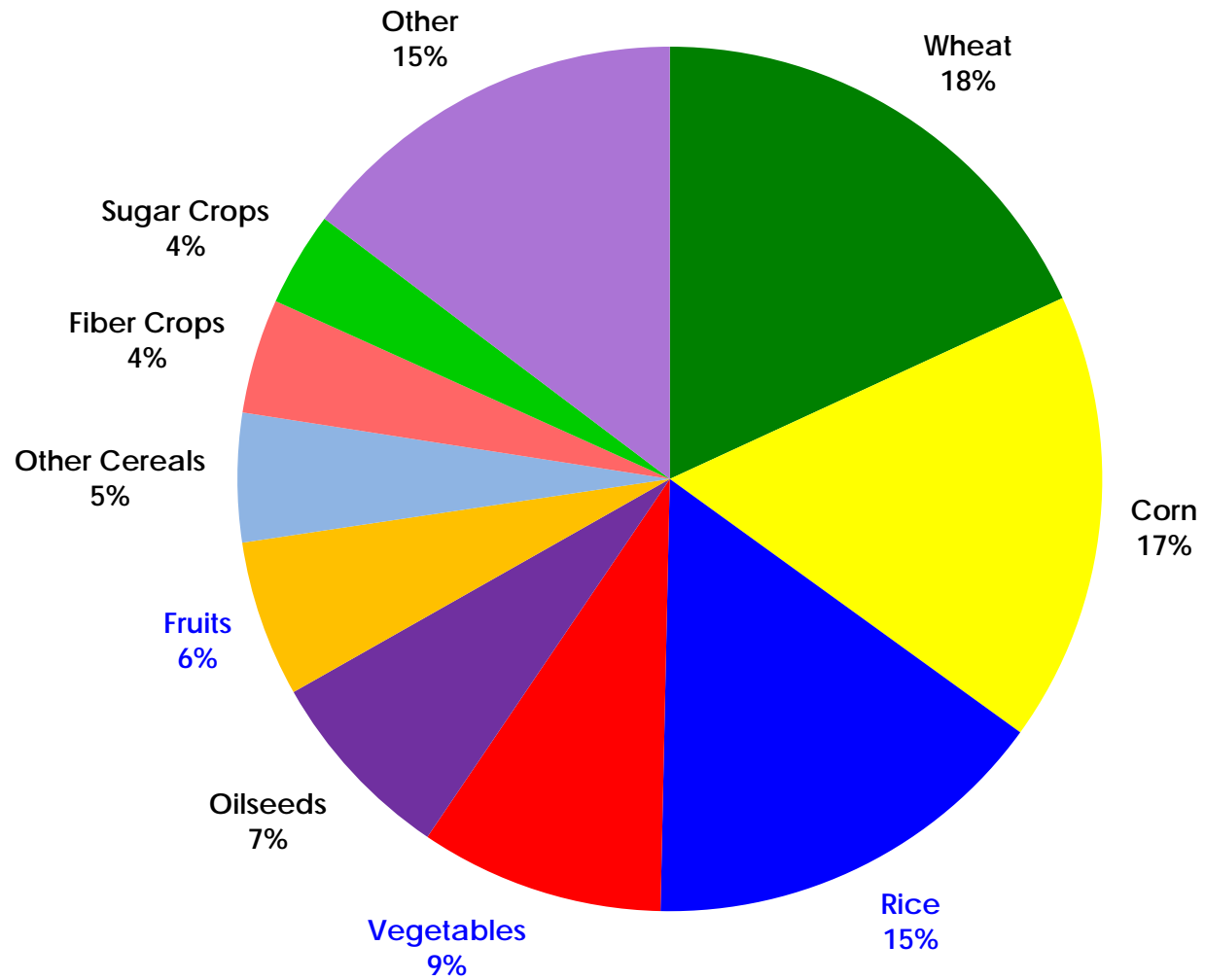


Global Ammonia Production



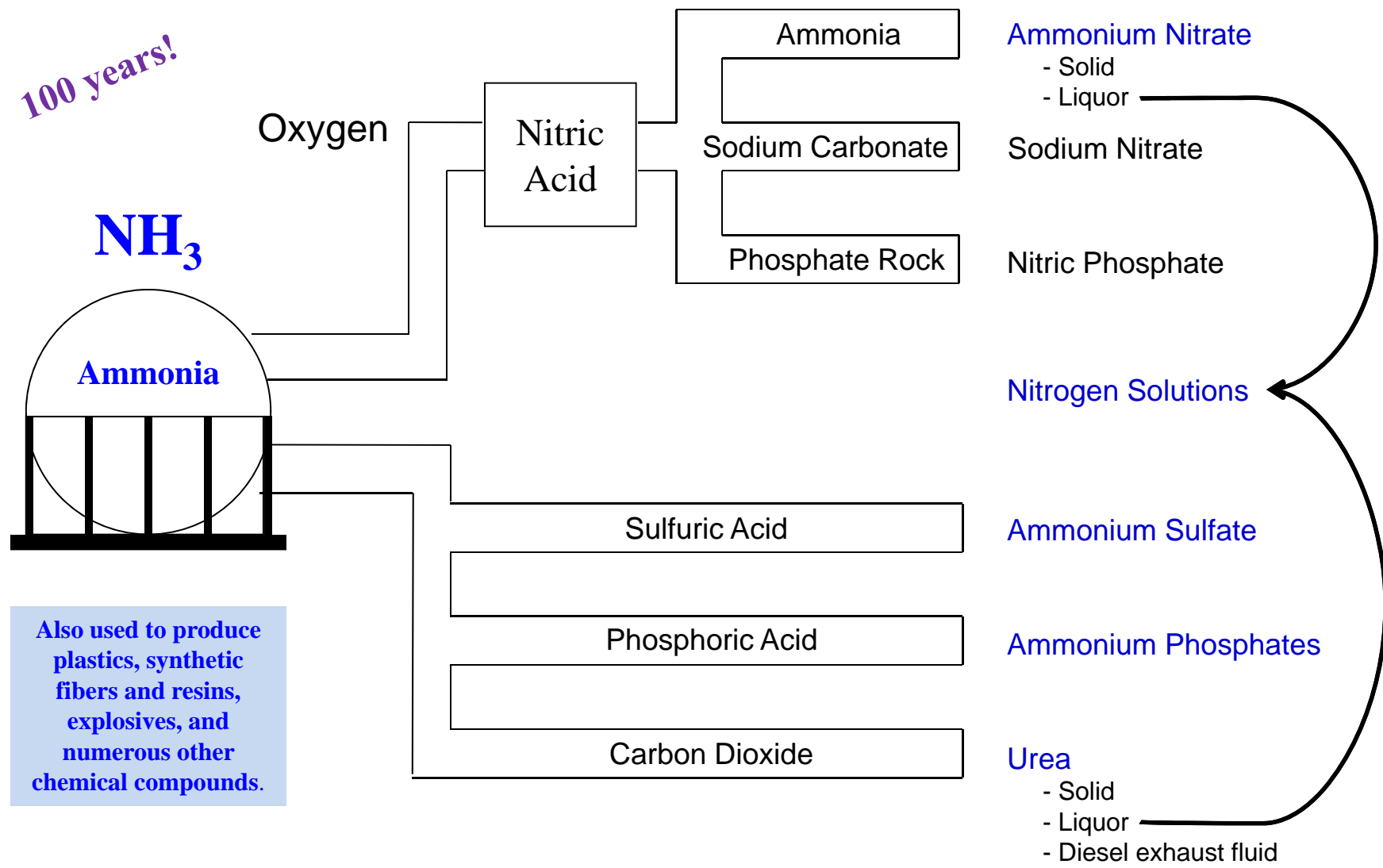
Source: 1943-1979: U.S. Geological Survey; 1980-2012: IFA; Smil, 2001.

Global Nitrogen Use by Crop, 2010/11



“Downstream” Nitrogen Fertilizers and Materials from Anhydrous Ammonia

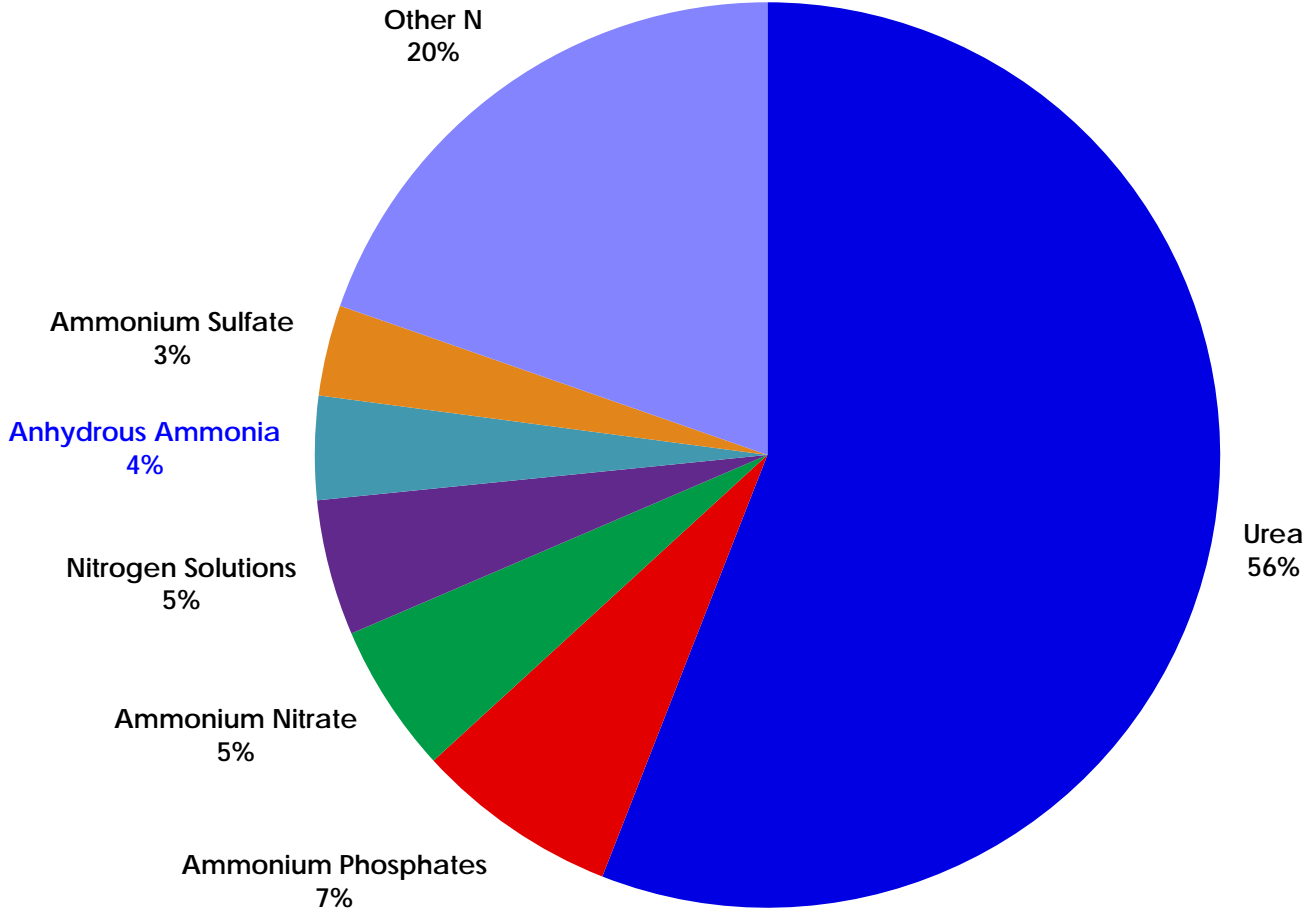
100 years!



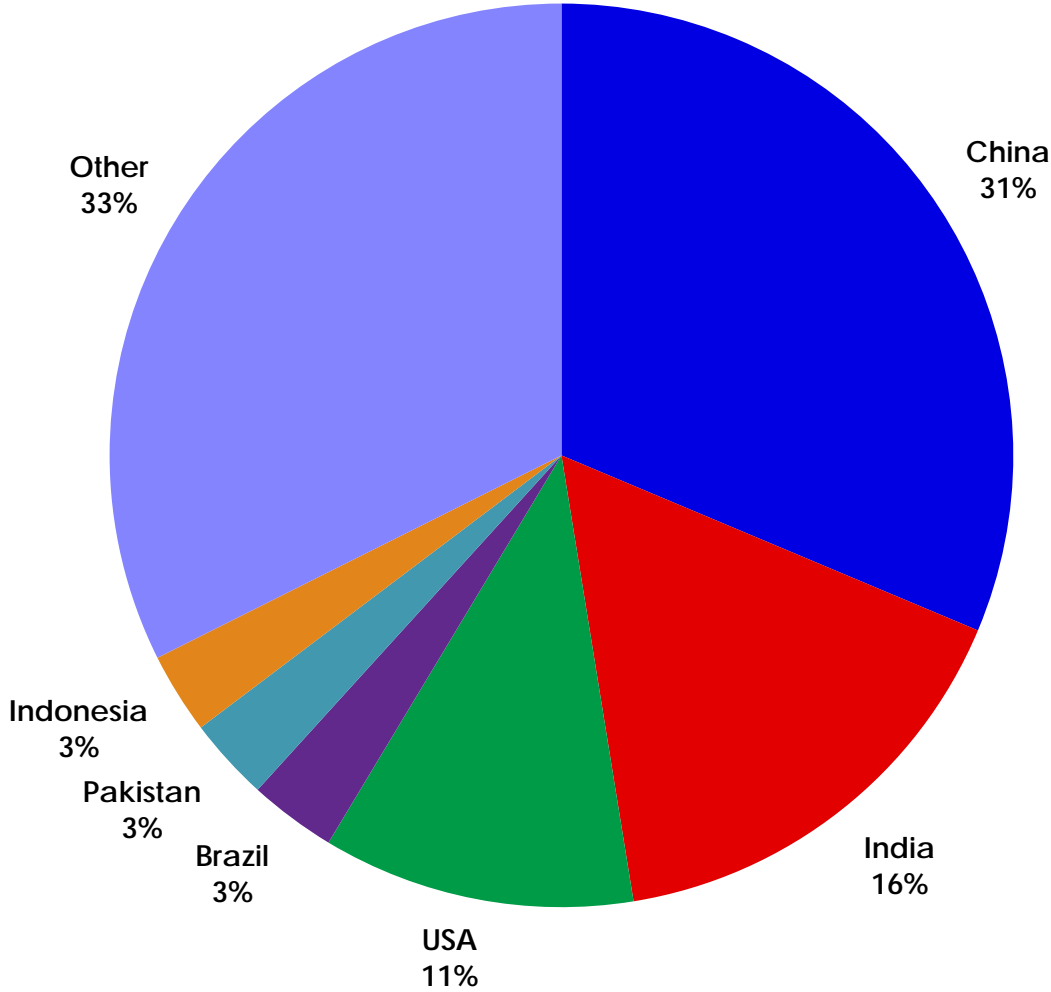
Also used to produce plastics, synthetic fibers and resins, explosives, and numerous other chemical compounds.

About 87% for fertilizer!

World Nitrogen Fertilizer Use 2011 - 107.9 million metric tons N

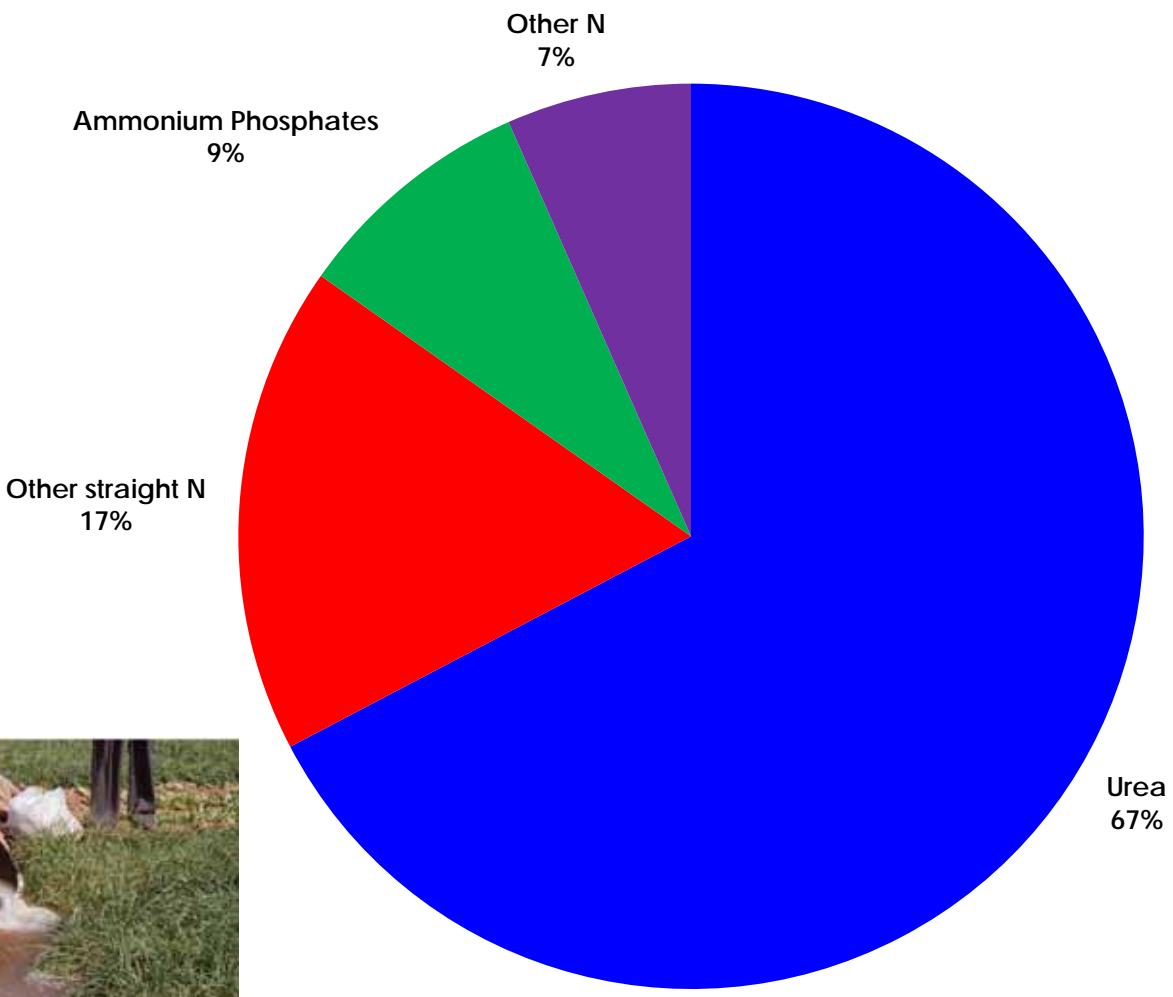


World Nitrogen Fertilizer Use 2011 - 107.9 million metric tons N



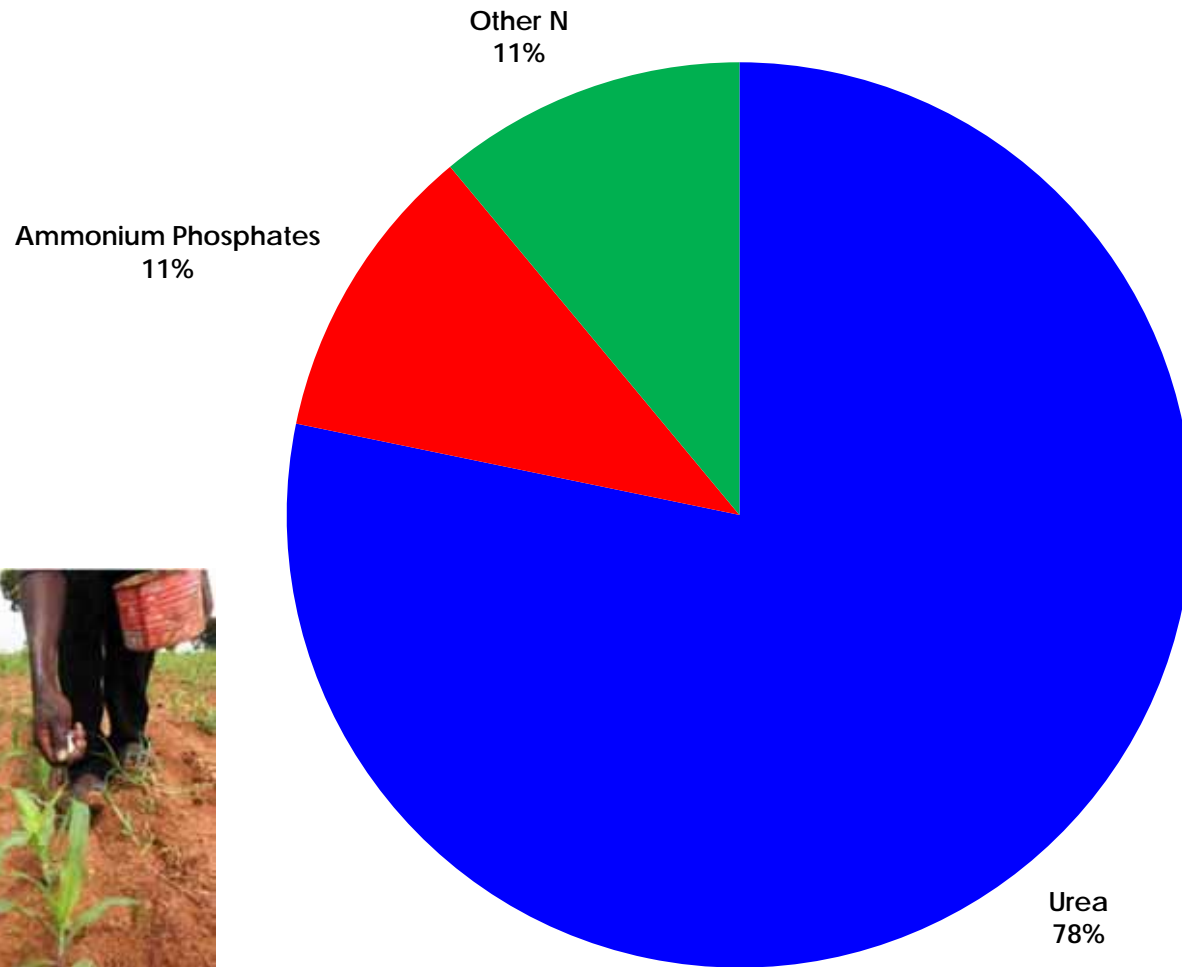
China - Nitrogen Fertilizer Use

2011 - 33.8 million metric tons N

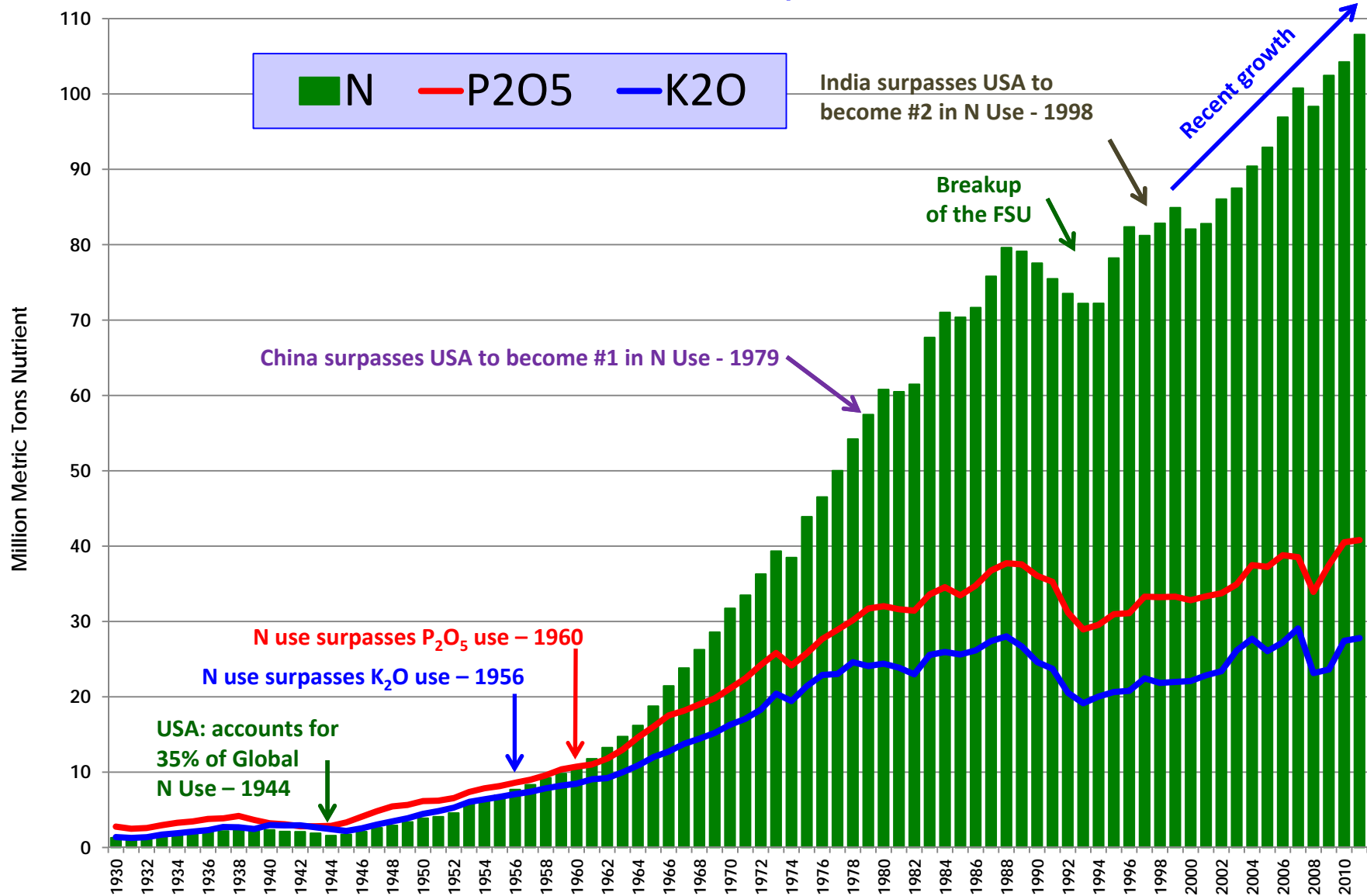


India - Nitrogen Fertilizer Use

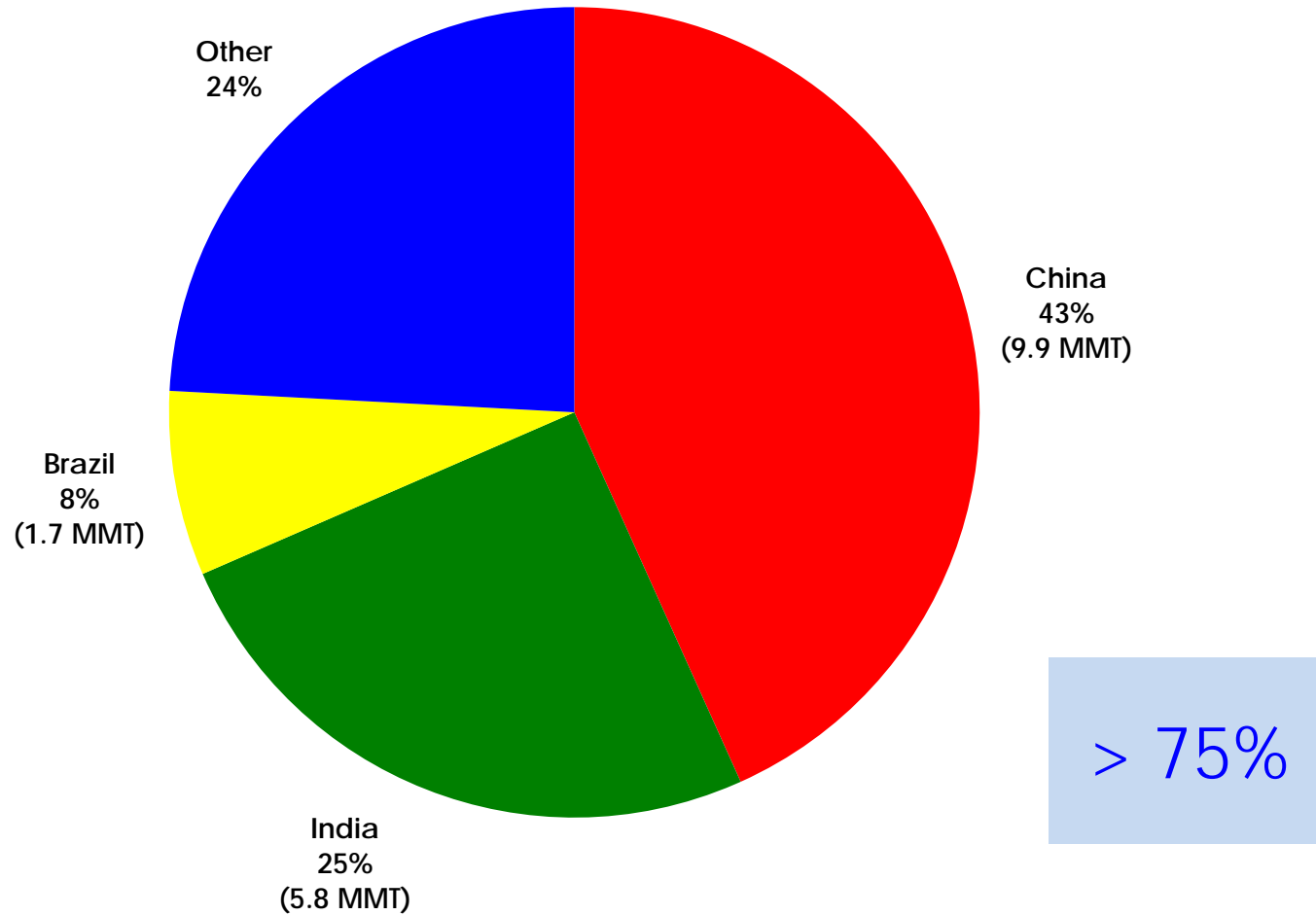
2011 - 17.4 million metric tons N



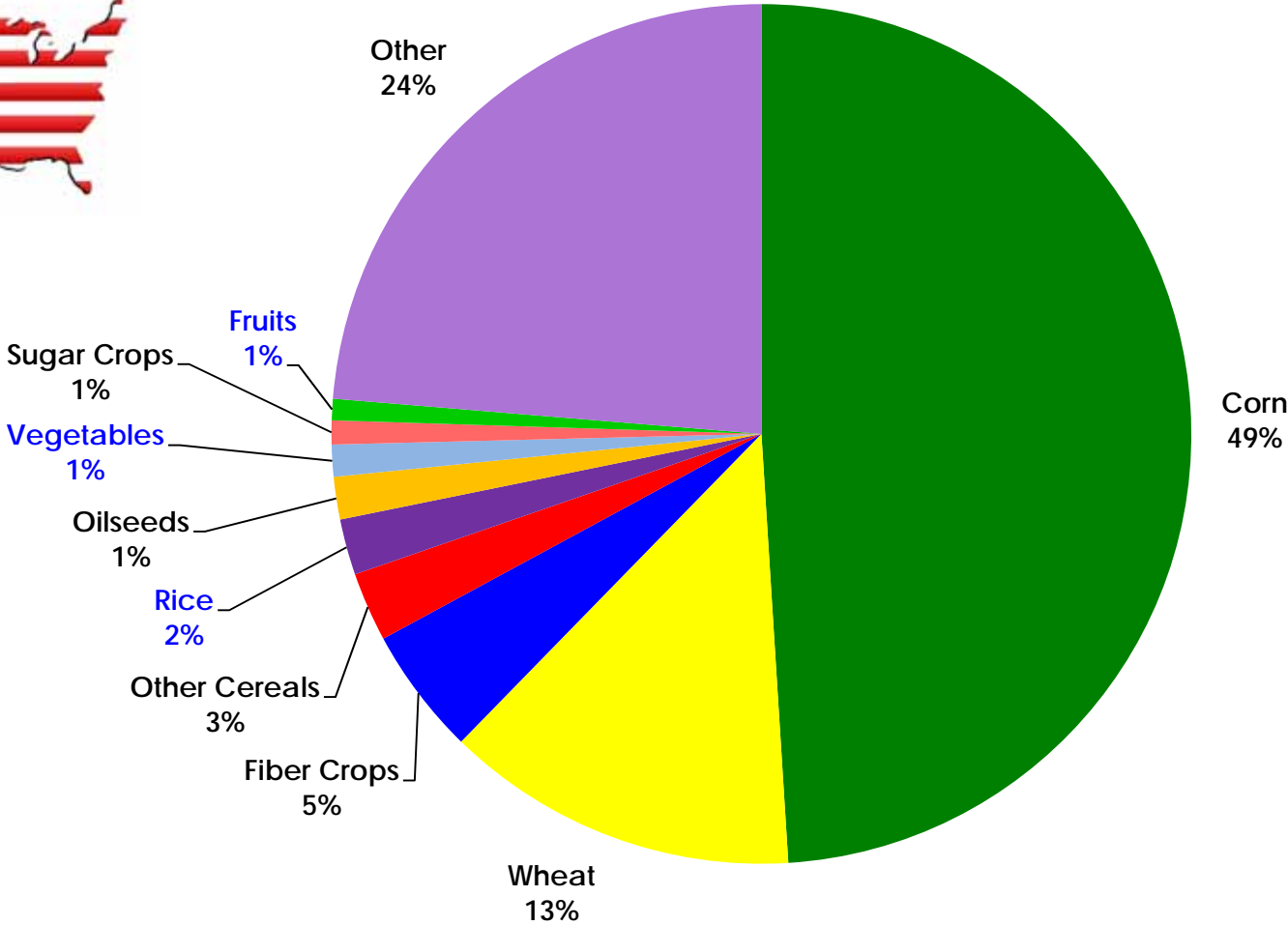
Global Nutrient Use, 1930-2011



Growth in Global Nitrogen Use 1999-2011: 23 MMT N

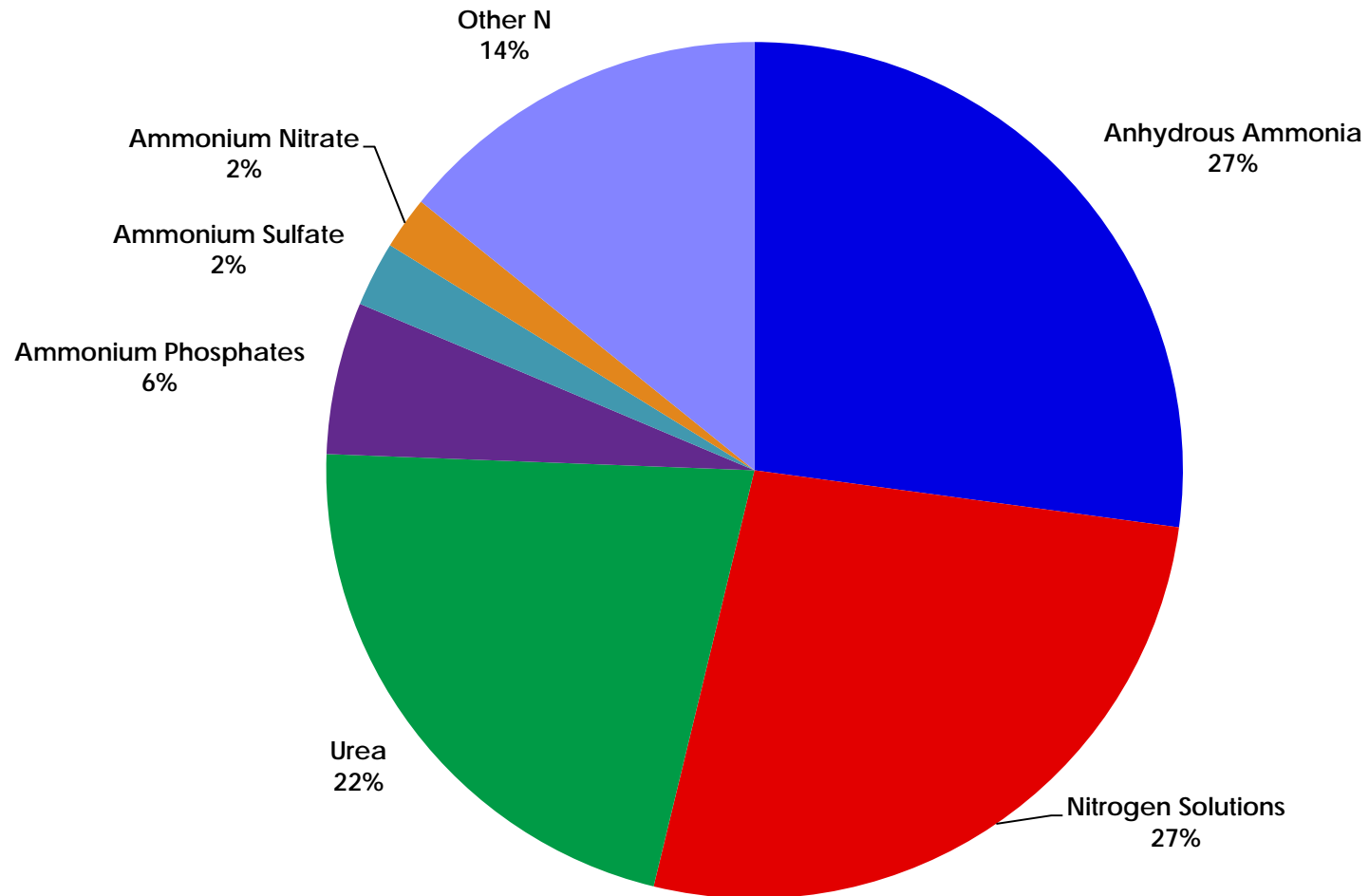


U.S. Nitrogen Use by Crop, 2010/11



U.S. Nitrogen Fertilizer Use

FY2010/11 - 12.84 million tons N



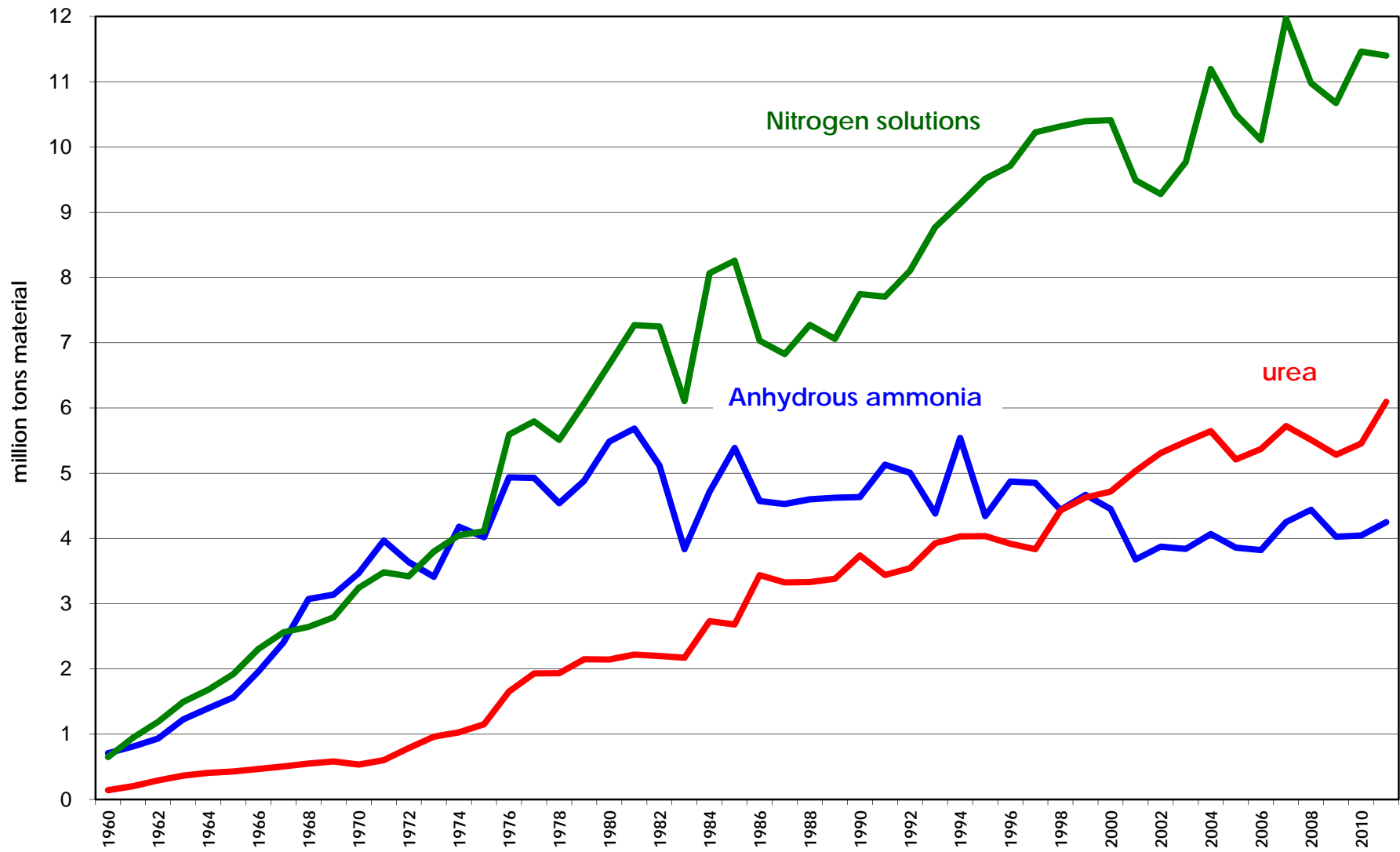


← Applying Anhydrous Ammonia

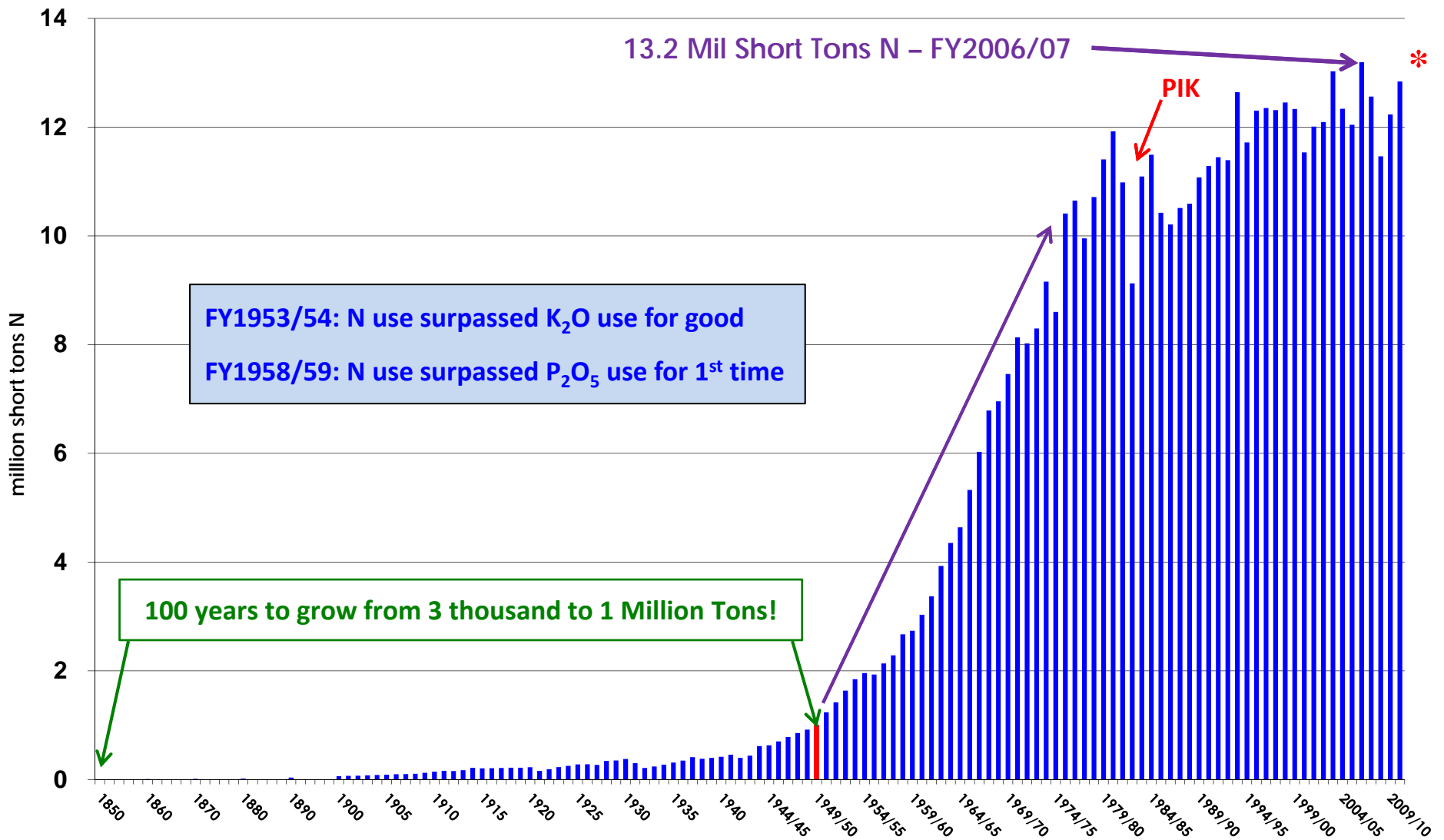
Side dressing nitrogen solutions →



U.S. Consumption of Selected Nitrogen Materials



U.S. Nitrogen Consumption: 1850 – 2010/11



Media Focus

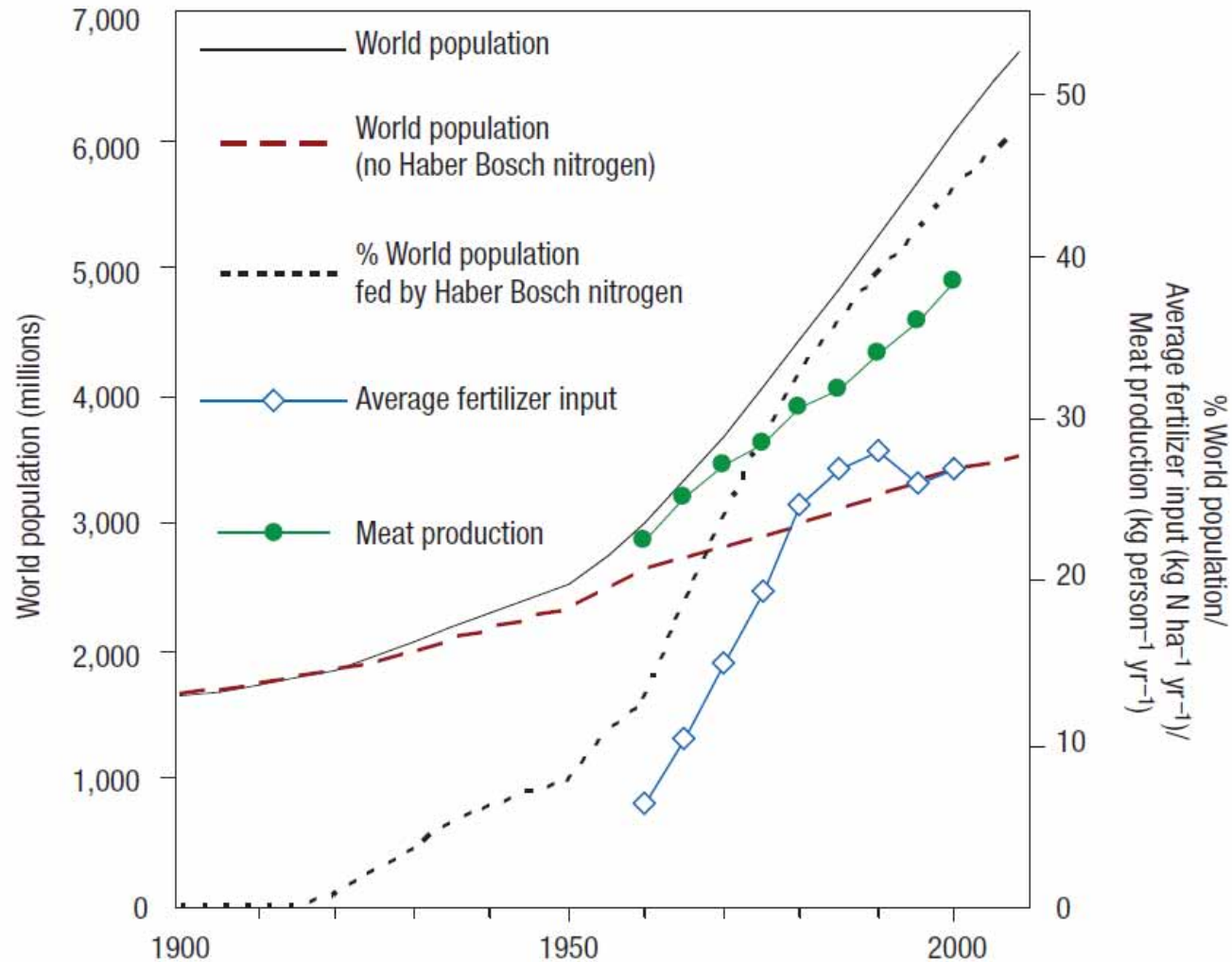


Here's My Plan to Improve Our World — And How You Can Help

Bill Gates, Wired.com, Nov. 12, 2013

“These days I get to spend a lot of time trying to advance innovation that improves people’s lives in the same way that fertilizer did.”

“... the lives of around half of humanity are made possible by Haber-Bosch nitrogen”



“The Haber-Bosch process has been of greater fundamental importance to the modern world than the airplane, nuclear energy, spaceflight or television”.

Vaclav Smil, *Enriching the Earth*, 2001.

Thank You!